

PHARMACOLOGICAL ASPECTS OF VITAMIN B₆

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The discovery and isolation of vitamin B₆ are closely related to the recognition of pellagra as an avitaminosis by Goldberger and Lillie (179). The disease is seen chiefly in the poorer populations of the southern parts of Europe and the Americas, whose food consists of a low-protein, corn-rich diet. Typical manifestations can also be induced experimentally by means of deficient diets. The "black tongue" of the dog has been considered analogous to the glossitis seen in man. The rat develops a localized dermatitis (acrodynia) on the paws, ears, and snout which is, to a certain extent, comparable with the characteristic skin lesions seen in man: dermatitis and browning of the skin, particularly in areas exposed to light. Further investigations revealed a relationship between the development of human pellagra and the low tryptophan content of zein, the protein of corn. This is of interest, since nicotinic acid and nicotinamide were recognized by Elvehjem *et al.* (142) as the "pellagra preventive factor" in man.

It is realized today that tryptophan can be metabolized in animal and man to nicotinic acid. While nicotinamide was established to be identical with the "anti-black-tongue factor" in the dog, another compound was recognized as the preventive or curative factor in the pellagra-like dermatitis of the rat (63, 193). "Adermin" was first prepared from yeast extracts by Kuhn and others and identified chemically as 2-methyl-3-hydroxy-4,5-bis(hydroxymethyl)py-

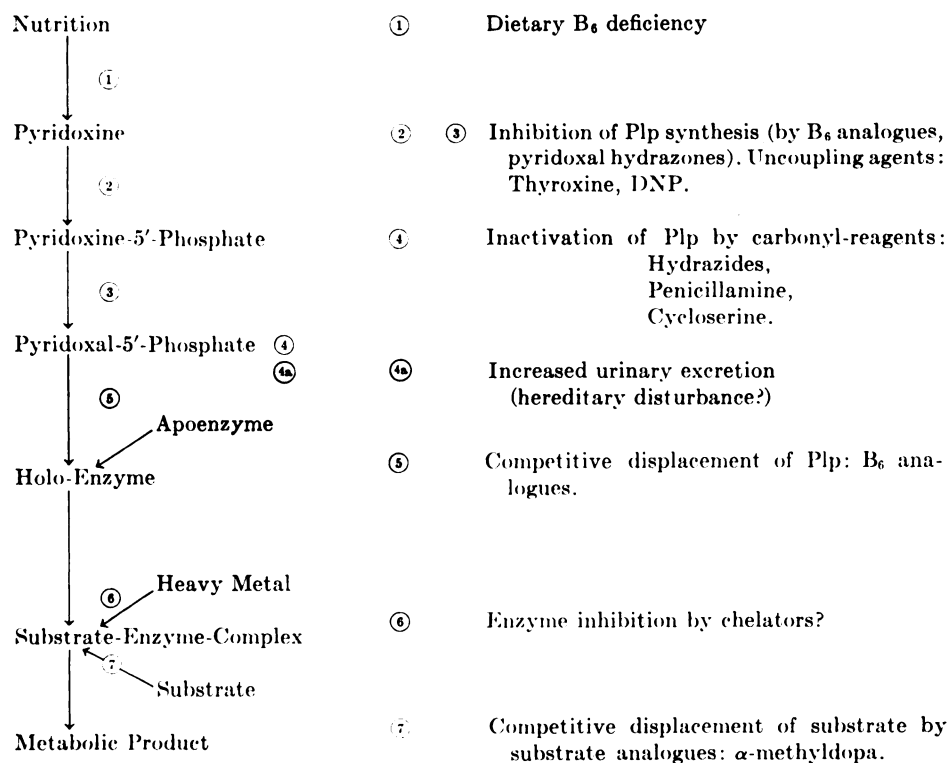


FIG. 1. Possible mechanisms of interference with the function of vitamin B₆.

ridine (for review see 193, 475, 539). As proposed by György and Eckhardt, it was called "pyridoxine" (194).

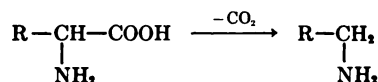
Until 1942 vitamin B₆ and pyridoxine were considered to be identical. In this year Snell *et al.* (477) found that the activity of "B₆ preparations" from animal material as assayed in microbiological tests varied up to a thousand times. What Snell then called "pseudopyridoxine" was later recognized to be a mixture of pyridoxal and pyridoxamine (27, 52, 164, 199, 215, 414, 415, 473, 475, 478).

Like most vitamins, the compounds of the B₆ group—pyridoxine (I), pyridoxal (II) and pyridoxamine (III) (see fig. 5)—exert their physiological effects after being phosphorylated to form coenzymes. Bound to specific proteins, they act as enzymatic catalysts in certain metabolic reactions. The most important are concerned with the metabolism of amino acids and amines (74, 76, 192, 475).

Although it is not yet possible to explain all physiological actions of vitamin B₆ in biochemical terms, it would be useful to begin with a review of the enzymatic reactions in which vitamin B₆ or its active form pyridoxal-5'-phosphate (Plp) (fig. 5, V) is involved as a coenzyme. The "pharmacological aspects" we will consider should be understood not primarily in terms of pharmacodynamic actions exerted by vitamin B₆ *per se*, but rather in terms of pharmacologically induced alterations of B₆-dependent physiological reactions, which result from a *functional* B₆ deficiency. A large number of drugs with well-defined pharmacological actions is known to promote "functional" B₆ deficiency by a variety of different mechanisms, as can be seen from figure 1: for instance by inhibition of coenzyme synthesis, by chemical inactivation of the coenzyme itself, or by competitive displacement of the coenzyme by structural analogues. A short review on B₆-dependent enzymatic reactions will therefore be confined to those, the disturbance of which could possibly explain the pharmacological action of the respective drug.

I. B₆ ENZYMES

A. Decarboxylation of aromatic amino acids



Vitamin B₆ is necessary for all nonoxidative metabolic reactions of amino acids. It was first recognized as a coenzyme of bacterial amino acid decarboxylases. Strains of *Streptococcus faecalis*, grown in B₆-free medium, lose most of their decarboxylase activity against tyrosine. The enzymatic activity is restored to normal by the addition of pyridoxal and ATP, from which Plp can be formed (52, 164, 192).

1. *3,4-Dihydroxyphenylalanine (dopa) and 5-hydroxytryptophan (5-HTP)*. In animals and man tyrosine is decarboxylated to tyramine not only by the intestinal bacterial flora but also by a tissue enzyme, detected first in kidney and liver (219). However, in the biosynthesis of the hormonal catecholamines tyrosine is not decarboxylated to tyramine, but hydroxylated to 3,4-dihydroxyphenyl-

alanine (dopa), which then is decarboxylated to dopamine by the Plp-dependent L-dopa decarboxylase. Dopamine is the immediate precursor of norepinephrine and epinephrine (for review see Holtz, 220).

It appears that L-dopa decarboxylase is identical with the enzyme that decarboxylates 5-hydroxytryptophan (5-HTP) to 5-hydroxytryptamine (serotonin) (230, 536). The suprarenal medulla and sympathetic nerves, which contain no serotonin, decarboxylate 5-HTP as well as L-dopa, while the serotonin-rich carcinoids of the intestine, which contain no catecholamines, decarboxylate L-dopa as well as 5-HTP. The decarboxylation of both amino acids is inhibited by α -methyl-dopa (536). There is, therefore, no direct correlation between the localization of the decarboxylating enzyme and the distribution of the biogenic amines.

Dopa and 5-HTP are not constituents of proteins but must be formed in the organism from the precursor amino acids tyrosine and tryptophan through the action of hydroxylases. The localization of the specific hydroxylases, in contrast to that of the nonspecific decarboxylase, then should be responsible for the specific distribution of the amines, dopamine, epinephrine, and norepinephrine in the one case, serotonin in the other. In brain, dopa decarboxylase activity and amine content run to a certain degree parallel; for example, both are higher in the brain stem than in the cortex (229, 511). However, the cerebral localization of the single amines is different (10, 511).

Dopamine seems to be not only an intermediary product in the synthesis of the catecholamines, but in certain parts of the body the final product of the biosynthesis and an active substance on its own. In the lung, as well as in the intestine and liver of dogs and cattle, dopamine is practically the only catecholamine to be found (147, 456). According to Sano *et al.* (446) as well as Carlsson (87) nearly all of the brain dopamine is found in the putamen and caudate nucleus, structures which are of importance for the functions of the extrapyramidal system (see section VI).

It is of interest that the injection of dopa as well as 5-HTP produces actions which may be potentiated by simultaneous injection of pyridoxine and which can be even more strongly enhanced by pretreatment of the animals with monoamine oxidase (MAO) inhibitors, which protect the resulting amines from inactivation. Intravenous injection of L-dopa in cats induced a rise of blood pressure and in rabbits hyperglycemia (222, 224); the injection of 5-HTP in guinea pigs caused bronchoconstriction (536). These actions were due to the amines arising from decarboxylation of the injected amino acids *in vivo*. They were potentiated by pyridoxine (31). After injection of dopa and 5-HTP central effects also are to be expected since these amino acids easily pass the blood-brain barrier and are then decarboxylated in the brain to the corresponding amines, dopamine and serotonin. After inhibition of brain MAO, *e.g.*, by previous injection of iproniazid, the injection of dopa or 5-HTP provoked central excitation and shortening of barbiturate and avertin (tribromethanol) sleeping time (221); the threshold for convulsions (electroshock and Metrazol, pentylenetetrazol) was raised (277) and the analgesic action of morphine increased (449, 454). Con-

comitantly the amine content of the brain was elevated. In contrast, reserpine causes a decrease in the amine content of the brain and at the same time sedation (*cf.* 80) and prolongation of barbiturate sleeping time (221); the seizure threshold is lowered (96) and the analgesic effect of morphine is abolished (449, 454).

2. *Histidine*. Histidine is decarboxylated to histamine not only by bacteria (for review see Gale, 163) but also by tissue extracts. In contrast to the bacterial enzyme, which decarboxylates L-histidine as well as D-histidine, the tissue enzyme decarboxylates only the L-form. Plp is the coenzyme. Histidine decarboxylase was first detected in kidney and liver (226, 530), and later in nervous tissue, with the highest activity in sympathetic nerves (229, 533). It is of interest that the noradrenergic nerves contain more histamine than norepinephrine, the actual transmitter substance (146, 533). A histidine decarboxylating enzyme, different from the histidine decarboxylase of the organs (rabbit kidney, guinea pig kidney and stomach, and fetal rat liver), is found in mast cells (166, 527). Both decarboxylases are Plp-dependent, but in contrast to tissue decarboxylase, the mast cell decarboxylase is not inhibited by α -methyl-dopa (418, 527, 529).

Since dopa and 5-HTP are decarboxylated by the same enzyme it was suggested by Udenfriend *et al.* (312, 501) that all aromatic amino acids, *e.g.*, dopa, 5-HTP, histidine, tyrosine (*o*-, *m*-, *p*-), tryptophan, phenylalanine, and α -methyl-dopa are substrates of a nonspecific "aromatic amino acid decarboxylase." However, Awapara *et al.* (25), working with a highly purified enzyme preparation from rat liver, found that only dopa, 5-HTP, and *o*- and *m*-tyrosine (but not *p*-tyrosine, histidine, and tryptophan) were decarboxylated. In agreement with this are results obtained by Hagen (195) with enzyme preparations from pheochromocytoma and by Holtz *et al.* (225) with kidney extracts of different species. Also Blaschko had found that *o*- and *m*-tyrosine in contrast to *p*-tyrosine were easily decarboxylated by the enzyme (67, 68b).

B. Decarboxylation of L-glutamic acid

The enzyme L-glutamic acid decarboxylase, which also has Plp as coenzyme, is found only in the central nervous system and catalyzes the decarboxylation of glutamic acid to γ -aminobutyric acid (GABA) (fig. 2). GABA was first detected by Ackermann and Kutscher (3) in 1910 in putrefying pancreas, and later by Roberts and Frankel (429) and Awapara *et al.* (24) in the CNS of various animal species. It is not found in cerebrospinal fluid, blood, or other organs (506) but it occurs in retina (283). The carbon skeleton of GABA is derived from glucose, as shown in experiments with brain slices and C¹⁴-glucose as substrate (53).

Roberts *et al.* (427, 428, 429, 431) found that the GABA concentration in brain is 4 to 5 times lower than that of glutamic acid. Highest concentrations of GABA and highest activity of glutamic acid decarboxylase (GAD) are found in the gray matter of brain (431), but none in peripheral sympathetic nerves or suprarenal medulla, which have particularly high dopa or 5-HTP decarboxylase activity (229, 230). Brain homogenates of small laboratory animals, such as

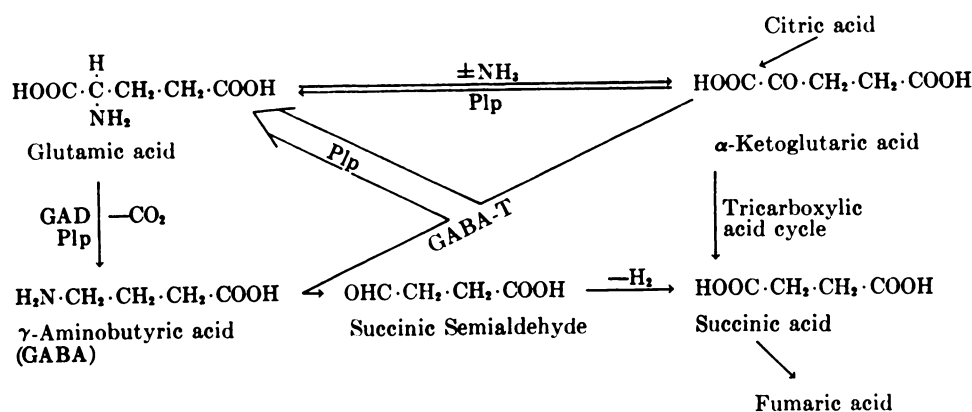


FIG. 2. Metabolism of glutamic acid and GABA in brain: GAD, glutamic acid decarboxylase; GABA-T, GABA- α -ketoglutaric acid transaminase.

mice, rats and guinea pigs, have higher GAD activities than homogenates from brains of larger species, *e.g.*, pig and beef (313). The relative activities found by Lowe *et al.* (313) were: in monkeys 100, in rabbits 187, in rats 251, and in mice 340.

GABA is easily taken up by brain slices, but not by slices of liver, kidney, or diaphragm (140, 447). In mouse brain the GABA content increases with age. GABA content and GAD activity parallel each other until the fifth postnatal day, while during further development GAD activity increases much more strongly than GABA content (430). This indicates that the disappearance of GABA by transamination (see fig. 2) keeps pace with its increased formation. That means an acceleration of turnover with increasing development, which maintains the GABA content at a constant level. Similar results were obtained with chick embryo and rabbit brain (428, 468).

GABA is a metabolite participating in the α -ketoglutaric acid–glutamic acid–succinic acid cycle, accessory to the tricarboxylic acid cycle and thereby involved in the cerebral carbohydrate metabolism (see fig. 2). GABA is also of pharmacological interest in that it exerts inhibitory actions on synaptic transmission in the CNS (for review see 139, 263, 428, 458, 499).

1. *GABA and GABA derivatives as central inhibitors.* Factor I, isolated by Florey (153) from brain and spinal cord of cattle, which proved to be an inhibitor of the stretch receptor of crustaceans, appeared to be identical with GABA (47). According to Kuffler and Edwards (293) GABA has an action on the stretch receptor of crayfish and lobster like that of the inhibitory neuron. It is unknown whether GABA itself is the transmitter substance. Therefore it is of interest that in mammalian brain GABA has inhibitory actions. From electrophysiological investigations in cats, Purpura *et al.* (410) have concluded that topical application of GABA to the cerebral cortex induces a blockade of excitatory synapses. According to Iwama and Jasper (245) GABA blocks selectively the superficial dendrites, either directly or through a blockade of synaptic

transmission. Grundfest (191) came to the conclusion that GABA does not activate inhibitory synapses, but inhibits excitatory synapses (139, 263).

It was pointed out that probably not γ -aminobutyric acid (GABA) itself but β -hydroxy- γ -aminobutyric acid (206) is the actual central inhibitor. This hydroxyamino acid, administered intracisternally in doses a tenth of those of GABA, suppressed convulsions induced electrically or by injection of Metrazol. It was detected by Ohara *et al.* (381) and by Hayashi (204, 205) as a brain constituent. On the other hand, Mitoma (360) found no transformation of C¹⁴-labeled GABA to β -hydroxy- γ -aminobutyric acid in brain homogenates.

In addition to acetylcholine, other choline esters such as butyrylcholine (212, 227), whose action is chiefly nicotinic, and γ -aminobutyrylcholine occur in mammalian brain (259, 295). The latter substance is said to have an inhibitory action 1000 times greater than that of GABA (493). GABA seems to be also the precursor of γ -guanidobutyric acid, which occurs in brain (243, 397) and probably originates from a transamidination between arginine and GABA, as demonstrated by Pisano and Udenfriend (398). In contrast to GABA, which inhibits the superficial excitatory synapses of the brain, the guanidino derivative is supposed to activate the inhibitory axodendritic synapses (411). On the stretch receptor the activity of this compound is one-third to one-seventh that of GABA (293).

The activity of GABA as an inhibitory modulator of neuronal activity can be shown only when it is topically applied to the brain cortex. It shows no detectable central action when injected intravenously because, under physiological conditions, it is unable to penetrate the blood-brain barrier in full-grown animals (171, 468, 500). The intravenous and intraperitoneal injection in mice and rats of as much as 3 g/kg did not increase the GABA content of the brain, and after the injection of 200 mg/kg intravenously in monkeys it was impossible to detect GABA in the cerebrospinal fluid (171). On the other hand, in younger animals, where the blood-brain barrier is not yet fully developed, an increase in the GABA content of the brain after intravenous injection can be found (468) as well as the typical reversal in the EEG of the surface-negative waves to surface-positive waves (409). Also, after local damage to the blood-brain barrier by craniotomy of long duration or application of ethyl chloride on a free circumscribed area of the cortex, the intravenous injection of GABA increased the GABA concentration within the damaged area (56) (see section VII).

2. *GABA as an intermediate in brain metabolism.* The GABA level in brain is the result of a dynamic equilibrium between formation and utilization. The most important enzyme for the utilization of GABA is the B₆-dependent GABA- α -ketoglutaric acid transaminase (GABA-T), which catalyzes the transamination of α -ketoglutarate to glutamate with the concomitant formation of succinic semialdehyde from GABA (see fig. 2) (for review see 6, 43, 139, 427, 428, 498, 499). It has been calculated from experiments with brain slices that this glutamate-GABA shunt contributes approximately 40% of the oxidative brain metabolism (344). Recently this figure was questioned by Haslam and Krebs (202a). Working with rat brain slices and homogenates they found that the glutamate-

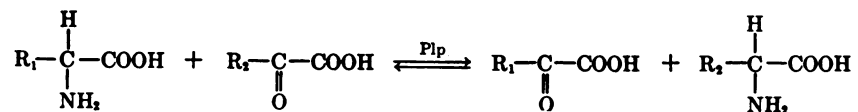
GABA shunt contributes only 10% to the cerebral oxidative metabolism. The oxygen consumption of cat brain slices can be maintained by GABA or succinic semialdehyde as well as by glucose or glutamate as substrates. Under these conditions the P:O ratio is between 2 and 3 (346).

It is of particular interest that the enzymes regulating the glutamate-GABA shunt, *i.e.*, GAD and GABA-T, are B₆-dependent. In contrast to GAD, GABA-T is found in subcortical brain areas in a much higher activity than in cortex. In all areas of the brain the ratio GABA-T/GAD activity is greater than 1 and it reaches values up to 50 in the inferior olivary nucleus and in the superior colliculus (6, 7, 444). The GABA content of the brain is more closely related to GAD than to GABA-T-activity. This can be altered by pharmacological agents which influence the two Plp-dependent enzymes to a different degree, because Plp has different affinities for the two apoenzymes (40, 43) (for details see V D,1).

The pH optimum for the decarboxylase (GAD) is about pH 6.5, while that of the transaminase (GABA-T) is pH 8.2 (432). At pH 7.5 both enzymes have the same percentage of their maximal activity. Therefore, a metabolically induced intracellular shift in pH to the acid side would enhance GABA formation with a concomitant increase in brain GABA, while a shift towards a higher pH would enhance utilization, thus lowering the GABA content.

The internal neuronal pH is mainly regulated by the CO₂-bicarbonate-buffer system. In the hippocampus with its particularly low seizure threshold, CO₂ can reduce spontaneous as well as experimentally augmented activity (135). This perhaps explains why inhibition of the carbonic anhydrase, *e.g.*, by acetazolamide, resulting in an increase in the cerebral CO₂ tension and thus reducing the intracellular pH, has an anticonvulsive action against electroshock in mice (185, 358). It could explain, too, why in cases of epilepsy acetazolamide suppresses the slow EEG waves provoked by hyperventilation (443).

C. Transaminases



These B₆-dependent enzymes are of basic importance for the formation of amino acids from nitrogen-free keto acids and thus for protein synthesis. Therefore, they are necessary for growth and cell replacement. The most important transaminases are those concerned with the transamination of glutamic acid, since they are capable of catalyzing the transfer of the NH₂ group of glutamic acid to almost any keto acid (for review see 348).

Formation of the transaminases mainly takes place in liver and muscle. Even under physiological conditions, small amounts of the intracellularly formed enzymes are released into the blood (422). An increased transaminase titer in serum is a criterion of cell injury, for example, after myocardial infarct or after liver injury produced by alcohol or carbon tetrachloride. An increased serum

titer can occur also with other enzymes, *e.g.*, lactic acid dehydrogenase, aldolase, and muscle creatine phosphokinase (for review see 422). Inhibitors of oxidative metabolism, such as CO, iodoacetate, fluoroacetate, and DNP, increase the release of intracellular enzymes into the blood (83).

The pattern of abnormal serum enzyme activity depends upon the enzymatic equipment of the affected organ (422). The most active transaminases found in human serum are glutamic-oxalacetate transaminase (GOT) and glutamic-pyruvate transaminase (GPT). GOT occurs in highest concentration in the heart, GPT in the liver. In cases of myocardial infarct, the serum GOT activity, almost exclusively, increases in proportion to the extent of necrosis. On the other hand, in acute toxic hepatitis (phosphorus, carbon tetrachloride) both transaminases are increased; the increase in serum GOT is even more pronounced than that of GPT. At the same time enzyme activities in the liver decrease. The determination of transaminase activity in serum is therefore of importance for clinical diagnosis (422).

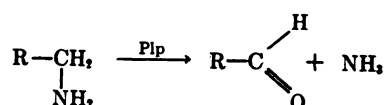
Activation of the anterior pituitary by stress causes an increase in the transaminase activity of various organs, chiefly in the liver (for review see 274), resulting from an augmented enzyme synthesis or an activation of enzyme precursors brought about by corticosteroids (274). Administration of corticoids (*e.g.*, prednisone, cortisone, or hydrocortisone) for several days induced hepatic GPT activity up to 13 times the initial values. Although the induced GPT activity could not be further stimulated by Plp, in B₆ deficiency hydrocortisone led to an increase of only about 50%, while, together with Plp, it was capable of increasing the activity by 600% (438). Since transaminases, especially GPT, are of importance also for hepatic gluconeogenesis, it is understandable that cortisone, which normally stimulates gluconeogenesis very strongly, produced only an insignificant increase in liver glycogen in B₆-deficient rats (138).

Racemization. It has been known for a long time that growth in the rat is promoted by D-amino acids as well as by L-amino acids. For example, L- and D-tryptophan are of equal value in the "growth test" (55). In experimentally produced anemia in the dog D-histidine and D-tyrosine were utilized to the same extent as the L-derivatives in the regeneration of hemoglobin (537). Since only the naturally occurring L-amino acids can be used for protein synthesis and growth, the conclusion was drawn that the unnatural D-forms were transformed by "racemization" to the natural L-forms. The experiments of Rose (435) have thrown light on the mechanism underlying "racemization." He found that keto acids were able to replace the corresponding L-amino acids in the growth test. For example, phenylpyruvic acid was equivalent to L-phenylalanine, and α -keto- γ -methyl-valeric acid to L-leucine or L-isoleucine. D-Amino acids are deaminated by D-amino acid oxidase to form optically inactive keto acids, which then are reaminated by optically specific transaminases to form L-amino acids. In agreement with this were results obtained after the administration of DL-dopa: in guinea pigs the amount of dopamine formed *in vivo* by L-dopa decarboxylase and excreted into the urine corresponded to the injected amount of the L-form, while in rats the amount of the urinary dopamine exceeded that which

could possibly have been derived only from the L-form (223, 366). This species difference was explained by the fact that rats have a much higher D-amino acid oxidase activity in liver and kidney than guinea pigs (287). Thus, in the rat D-dopa could be quantitatively deaminated to dihydroxyphenylpyruvic acid; the latter then was transaminated to L-dopa, the substrate of L-dopa decarboxylase.

Thus, also in this way, vitamin B₆ is indirectly involved in the racemization of amino acids in animals. True racemases dependent on vitamin B₆ have so far been detected only in bacteria, not in animal tissues.

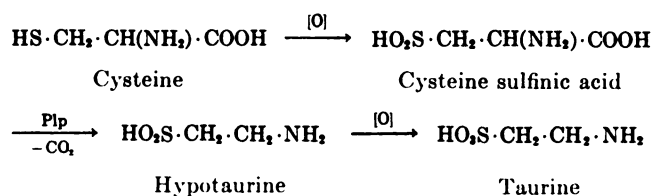
D. Amine oxidases



These enzymes are of particular interest because they inactivate and thereby detoxify some pharmacologically active amines by oxidative deamination. First indications that Plp may be a coenzyme were given by observations that histaminase was inhibited by carbonyl reagents (120, 534, 567). Electrophoretically pure histaminase preparations from hog kidney, which deaminated histamine, but not cadaverine and putrescine, contained Plp and flavin (254, 255). Thus, B₆-dependent enzymes—histidine decarboxylase (see section I A, 2) and histaminase—catalyze synthesis as well as breakdown of histamine.

Similarly, the monoamine oxidase of plasma was presumed to have Plp as coenzyme because of the inhibitory action of carbonyl reagents (492). Yamada *et al.* have recently shown that the crystalline enzyme which acts only on aliphatic amines such as spermine, butylamine, amylamine, and heptylamine, but not on aromatic amines like tyramine, norepinephrine, tryptamine, and serotonin, contains a Cu⁺⁺(Plp)₂-complex (557, 558). The monoamine oxidase of liver, kidney, intestine and brain, which deaminates the aromatic monoamines, is not B₆-dependent (for review see 400).

E. Enzymes of the metabolism of sulfur-containing amino acids

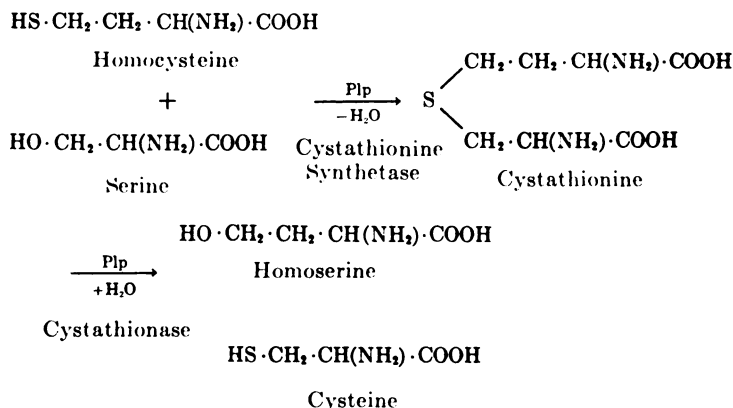


As first shown by Blaschko *et al.* (66, 68) cysteic acid is decarboxylated by a B₆-dependent enzyme in the mammalian liver. Because cysteine sulfinic acid is more rapidly decarboxylated by the same enzyme, this is the most likely main pathway of taurine formation (25a, 55a, 68c, 230a).

The enzyme in brain and liver is very sensitive to a dietary B₆ deficiency. This

explains why taurine excretion in urine ceases in a B₆-deficient state and reappears when pyridoxine again is added to the diet (55a, 68, 68a). However, in spite of a diminished excretion into the urine, the taurine content in liver (4) and in brain (230b) remains normal, while it is even increased in spleen and muscle (4).

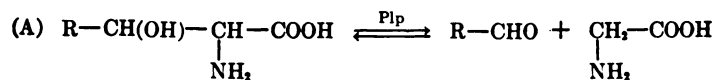
Biosynthesis of cystathionine from homocysteine and serine (cystathionine synthetase) is also B₆-dependent, as is its breakdown by cystathionase to homoserine and cysteine (61a; for review see 76):



In B₆ deficiency the urinary excretion of cystathionine is increased (230c). This could be explained by a higher sensitivity of cystathionase toward B₆ deficiency than of cystathionine synthetase. Cystathioninuria in man has been reported in two adult patients with mental aberrations (160a, 197a). Although no signs of B₆ deficiency could be observed, one of these patients responded to administration of pyridoxine with a reduction of cystathionine excretion and a concomitantly enhanced sulfate excretion. Other vitamins were ineffective (160a). However, there seems to be no causal relationship between this presumably "inborn error of metabolism" and mental disorders, since other members of the two families who were mentally normal displayed the same metabolic disorder (160a, 197a).

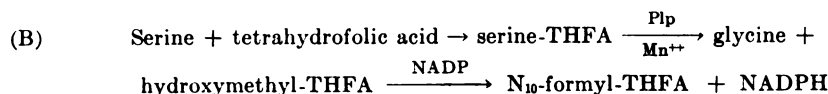
Concerning the metabolism of other sulfur-containing amino acids it is of interest that not only transsulfuration from homocysteine to cysteine (as seen above), but also desulfuration of methionine, cysteine, and cysteine sulfinic acid is catalyzed by B₆-containing enzymes (for review see 74, 75, 76).

F. Serine and threonine aldolase

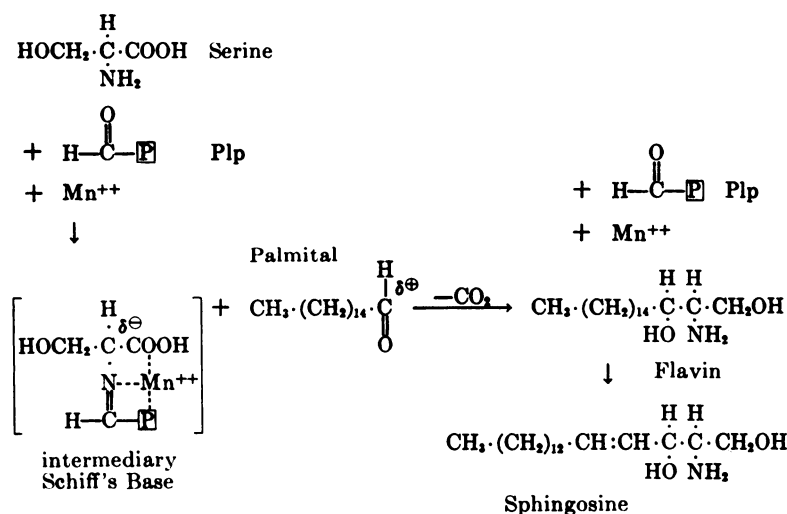


α -Amino- β -hydroxypropionic acid (serine) and α -amino- γ -hydroxybutyric acid (threonine) are metabolized to glycine and the corresponding aldehyde by B₆-dependent aldolases. The formaldehyde formed from serine, in contrast to the acetaldehyde formed from threonine, is coupled to tetrahydrofolic acid

(THFA). The resulting hydroxymethyl-tetrahydrofolic acid is then oxidized by NADP (formerly called TPN) to N_{10} -formyltetrahydrofolic acid. This compound is the biological carrier for one-carbon moieties and is of importance for the synthesis of purines and pyrimidines (8, 203, 249; for review see 75).



The reaction in (A) from right to left is of importance for the synthesis of 1) δ -aminolevulinic acid, the precursor of heme (see section IX), and 2) the unsaturated amino alcohol sphingosine, a constituent of sphingomyelin. This synthesis most likely results from the reaction between serine and palmitaldehyde in which first serine forms a Schiff's base with Plp and Mn^{++} as chelator. This complex undergoes an aldol condensation with palmital with concomitant decarboxylation. The aldol is then reduced to sphingosine by a flavin-containing enzyme (72, 73, 486).



G. Enzymes of tryptophan metabolism

As can be seen from figure 3, several of the enzymes of the tryptophan metabolism require Plp as coenzyme. The B_6 -dependent tryptophan synthetase, which forms tryptophan (I) from indole (II) and serine, as well as tryptophanase, which metabolizes tryptophan to indole (II), ammonia, and pyruvic acid, are found only in bacteria (504, 555). All other metabolic pathways are found in animals and man as well as in bacteria. The B_6 -dependent decarboxylation of tryptophan to tryptamine (III) (532) as well as the decarboxylation of 5-hydroxytryptophan (IV) to serotonin (V) (102) is quantitatively negligible. The chief metabolic pathway is the cleavage of the indole ring by tryptophan pyrrolase to an unstable N-formylkynurenine intermediate (276, 284), which in turn is converted to kynurenine (VI) and 3-hydroxykynurenine (VII). These two compounds can be

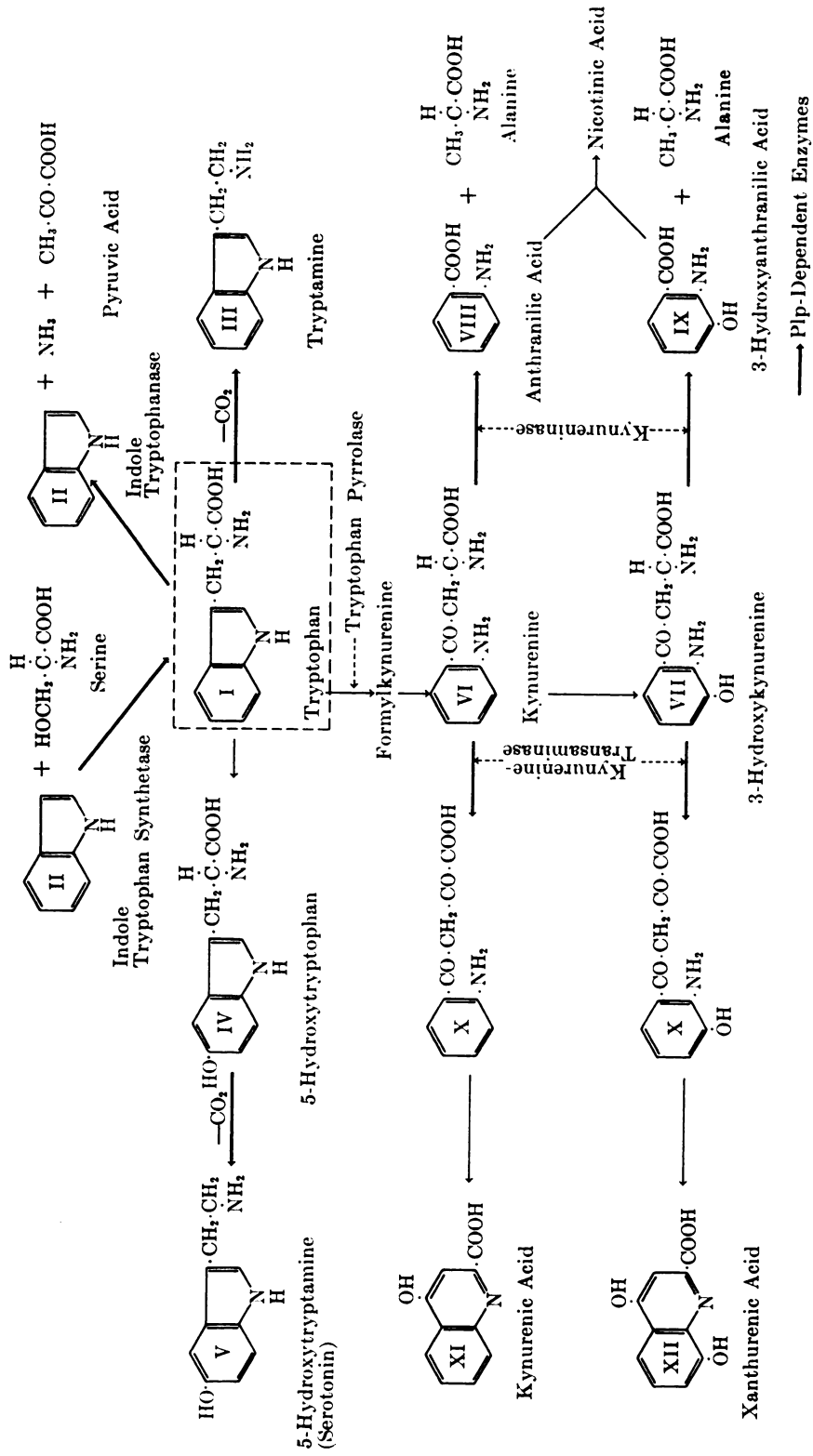


Fig. 3. Metabolism of tryptophan.

further metabolized by two B₆-dependent enzymes: 1) by kynureninase to anthranilic acid (VIII) or to 3-hydroxyanthranilic acid (IX), respectively; and 2) by kynurenine transaminase *via* the corresponding keto acids (X) to form kynurenic acid (XI) or xanthurenic acid (XII), respectively (75, 76, 81, 174).

In vitamin B₆ deficiency there is an inhibition of these two enzymes, and therefore an augmented excretion of kynurenine (VI) and 3-hydroxykynurenine (VII) as well as their acetylated derivatives (114). Kynurenic acid (XI) and especially xanthurenic acid (XII), which normally appear only in trace amounts in the human urine (405), are excreted in large amounts (305, 402). The "xanthurenic acid excretion test," *i.e.*, the abnormal increase of urinary xanthurenic acid after loading with L- or DL-tryptophan, is clinically used to detect vitamin B₆ deficiency (271, 272, 273, 304, 324, 512, 513). This paradoxical phenomenon, that a metabolite, xanthurenic acid, is formed and excreted in larger amounts than normal in spite of a lack of the coenzyme necessary for its formation, can be explained in the following manner.

Under physiological conditions the formation of formylkynurenine by tryptophan pyrrolase is the rate-limiting step in the degradation of tryptophan. Thus, even in B₆ deficiency, xanthurenic acid excretion is not necessarily increased. However, when a tryptophan load is imposed, this means not only an increased amount of substrate to be metabolized, but also a substrate-induced stimulation of tryptophan pyrrolase activity. Tryptophan pyrrolase activity is no longer rate limiting, and the conversion of kynurenine and 3-hydroxykynurenine to anthranilic acid and 3-hydroxyanthranilic acid, respectively, by the B₆-dependent kynureninase becomes the rate-limiting reaction. The formation of kynurenic acid and xanthurenic acid by transamination does not become rate-limiting because of the different affinities of the two apoenzymes for Plp: kynureninase is more strongly affected by B₆ deficiency than transaminase; therefore, the overflow of tryptophan metabolites occurs *via* the transamination pathway, leading to an increased formation and excretion of xanthurenic acid. Not yet explained is the fact that kynurenic acid, though formed by the same transaminase as xanthurenic acid, is not necessarily excreted in larger amounts, as is true for xanthurenic acid (for review see 13, 274).

However, the "xanthurenic acid excretion test" is not always positive in B₆ deficiency. Thus, hydrazides (*e.g.*, isonicotinic acid hydrazide, hydroxylamine, hydrazine, and isopropylhydrazine) which form hydrazones with Plp, thereby blocking its coenzyme function, even in convulsive doses did not increase xanthurenic acid excretion in rats after a tryptophan load (32, 436). On the other hand, subconvulsive doses of thiosemicarbazide and of thiocarbohydrazide as well as of the antimetabolite deoxyribose caused an abnormally high xanthurenic acid excretion (32, 543). In the dog there was an increased excretion of xanthurenic acid after the administration of INH, but not after deoxyribose (545).

H. Pyridoxal-5'-phosphate as a cofactor of phosphorylase

This enzyme, which contains Plp (33, 257), catalyzes the breakdown of glycogen by splitting off glucose-1-phosphate. Baranowski *et al.* (33), working with a

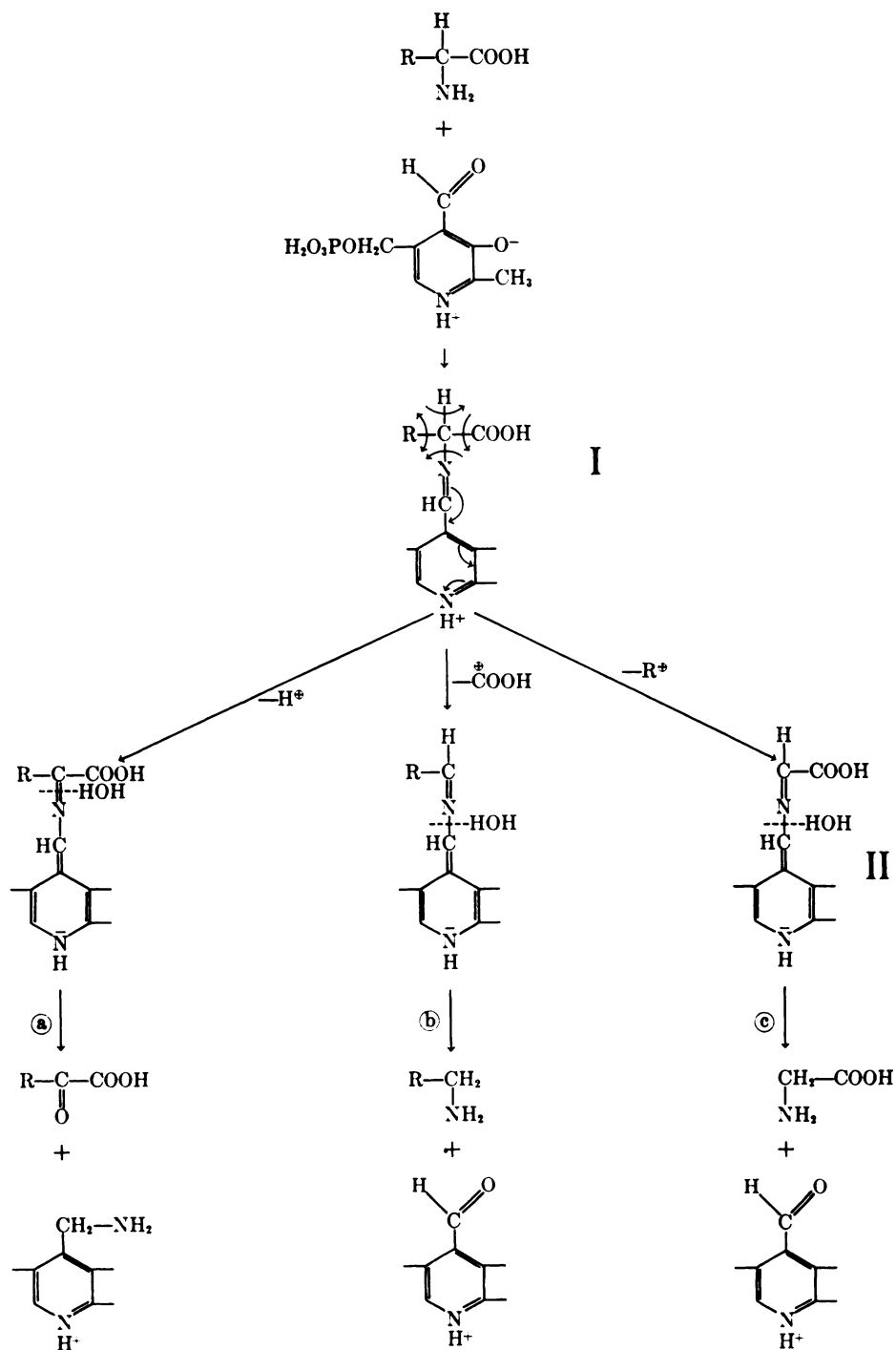


FIG. 4. The general mechanism for reactions catalyzed by pyridoxal-5'-phosphate.

crystalline muscle phosphorylase "a" preparation were able to isolate Plp as a barium salt. It is of interest that this was the first time that Plp was isolated from a natural source. The phosphorylase is the only B₆-containing enzyme not involved in the metabolism of the amino acids. Although the chemically induced dissociation of Plp from the purified enzyme results in an inactivation of the latter, B₆ presumably has not the function of a coenzyme (241). It seems to be necessary for the stabilization of the enzyme structure. In B₆ deficiency the total activity of muscle phosphorylase is diminished but the glycogen content of the muscle is unaffected, presumably because the activity of phosphorylase "a," the active form of the glycogen-metabolizing enzyme, remains unaltered (242, 315).

I. On the mechanism of reactions catalyzed by pyridoxal-5'-phosphate

In order to understand certain pharmacological actions which can be referred to the influence of a drug on B₆-catalyzed enzymatic reactions, it is necessary to have some knowledge of the biochemical mechanisms by which Plp participates in these reactions. Braunstein (74) and Snell *et al.* (352, 475) have shown that almost all of these reactions proceed by the same mechanism, whereby pyridoxal, phosphorylated in the 5' position, is the most important catalytic form of vitamin B₆. Even in the absence of the apoenzyme, decarboxylation, transamination, and racemization of amino acids can be catalyzed by Plp or pyridoxal itself, if heavy metals are added (352). The phosphorylated 5-hydroxymethyl group, together with the pyridine nitrogen, is important for the binding of the coenzyme on the apoenzyme, but does not take part in the actual chemical or enzymatic reaction. The free aldehyde group in the position *para* to the pyridine-nitrogen as well as the free phenolic group in the position *ortho* to the aldehyde group are essential and participate in these reactions. Amino acids require a free amino group in order to react with Plp. N-Alkylated amino acids for instance are not decarboxylated.

As can be seen from figure 4, the first step in every Plp-catalyzed reaction is the formation of a Schiff's base (I) or of transitional Schiff's bases (II) (214, 352, 396, 452, 531). This results in a labilization of all four bonds around the α -C-atom of the amino acid (352, 396). Depending upon the substrate affinity and the apoenzyme specificity there then follows: a) transamination, which consists in the formation of the corresponding keto acid with the splitting off of H⁺ while the NH₂-group becomes coupled to pyridoxal to form pyridoxamine; b) decarboxylation, *i.e.*, cleavage of the carboxyl-carbon— α -carbon bond and the formation of the corresponding amine plus CO₂; or c) splitting off of the alkyl residue R with the formation of a lower homologue of the reacting amino acid (*cf.* section I F above).

II. METABOLISM OF VITAMIN B₆

All forms in which vitamin B₆ occurs in the organism (see fig. 5)—pyridoxine (I), pyridoxal (II) and pyridoxamine (III)—are practically equivalent because they can be enzymically converted into one another (74, 475, 476). Pyridoxine is

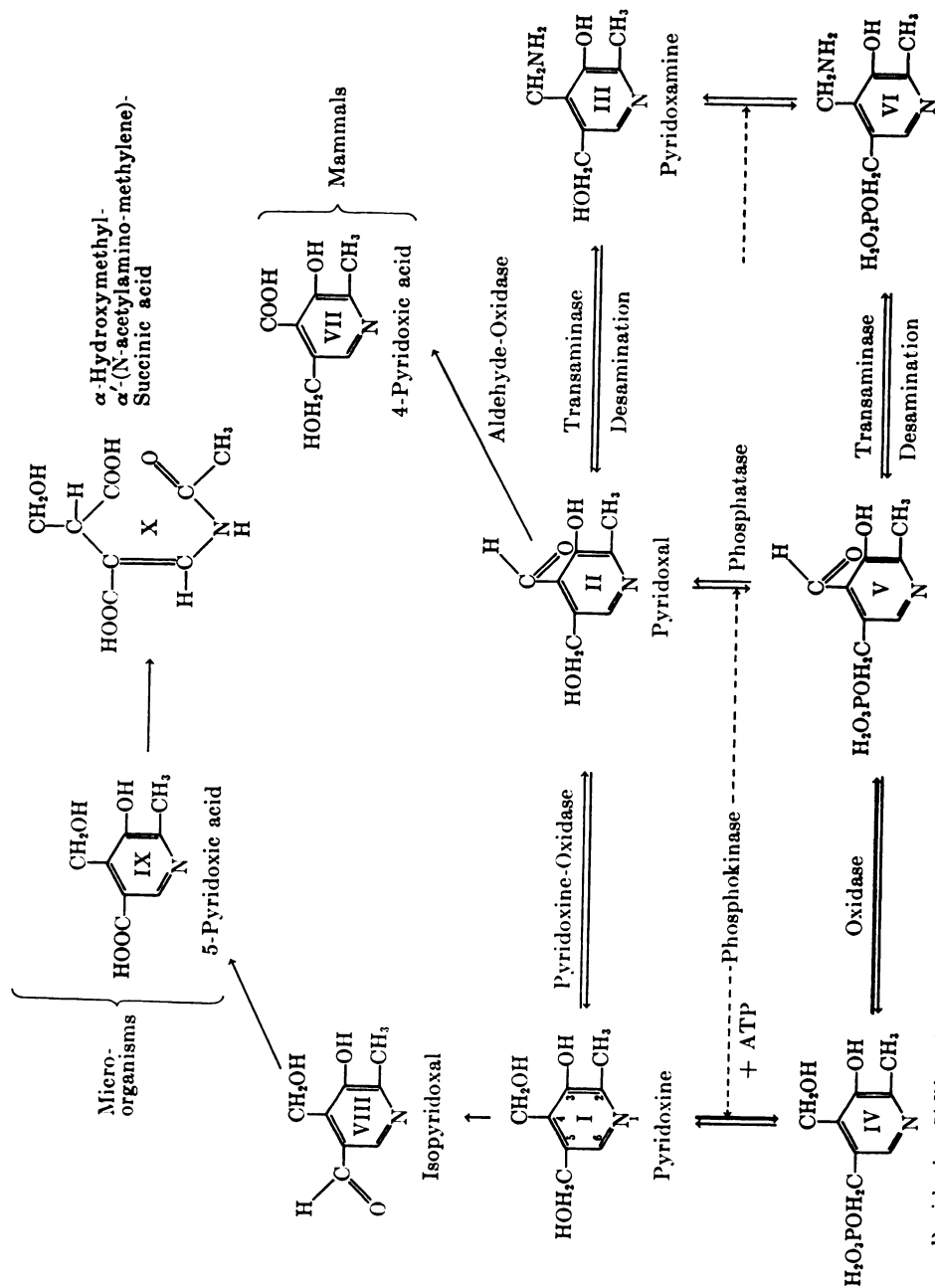


Fig. 5. Forms of vitamin B₆: Synthesis and degradation of pyridoxal-5'-phosphate. → Main pathway of Plp-synthesis.

oxidized by an oxidase to pyridoxal (519) which can be further transaminated to form pyridoxamine. All three forms can be phosphorylated in position 5' by an ATP-dependent phosphokinase (335). The oxidation of pyridoxine to pyridoxal presumably is preceded by phosphorylation (335, 519). Vitamin B₆ occurs in the tissues, *e.g.*, brain and liver, predominantly in the phosphorylated forms, Plp (V) and pyridoxamine-5'-phosphate (VI) (28, 314). The degradation of the phosphorylated forms is initiated by a phosphatase. In the case of pyridoxine-5'-phosphate (IV), the resulting pyridoxine (I) is oxidized to pyridoxal (II); in the case of pyridoxamine-5'-phosphate (VI) the resulting pyridoxamine (III) is deaminated to pyridoxal (II). Pyridoxal is then oxidized by a nonspecific aldehyde oxidase to pyridoxic acid (VII). This enzyme is found in liver, kidney, and other tissues. Pyridoxic acid is the only excretion product of metabolized vitamin B₆ (304, 476).

In bacteria the hydroxymethyl group of pyridoxine in position 5, but not that in position 4, is oxidized to form isopyridoxal (VIII), which is subsequently further oxidized to 5-pyridoxic acid (IX). Further degradation results in the cleavage of the ring to form α -hydroxymethyl- α' -(N-acetylamino-methylene)-succinate (X) (240, 434).

III. ABSOLUTE AND RELATIVE B₆ DEFICIENCY

While the rat is able to meet a considerable part of its B₆ requirement through bacterial synthesis in the intestine (141, 353), it appears that man is completely dependent upon the oral intake of the vitamin (111, 304, 488, 509). In view of the ubiquitous distribution of B₆ in all foodstuffs a mixed diet is generally sufficient to fulfill this requirement. The human B₆ requirement is reported to be 2 to 3 mg per day (304, 509). The increasing use of chemical and physical methods for the preparation and preservation of foodstuffs can, however, endanger the fulfillment of the vitamin B₆ requirement. The storage as well as the sterilization of meat and milk products by heat or radiation can lead to a significant loss of B₆ activity, particularly of pyridoxal and pyridoxamine (123, 417, 421). On the other hand, pasteurization or drying of milk causes only slight loss of vitamin activity (237, 556).

In 1952 and 1953 "epidemic B₆-deficiency symptoms" occurred in the USA caused by the exclusive use of a vitamin B₆-poor milk preparation (Liquid SMA, liquid synthetic milk adapted) in the nutrition of babies (59, 109, 332, 362). The most marked signs were epileptiform convulsions, and these could be promptly relieved by the administration of pyridoxine. Because of the relatively low B₆ content of human milk, 130 μ g/l (488, 556), the occurrence of convulsions is possible also in breastfed babies. Indeed, Bessey *et al.* (59) have reported two such cases, in which the convulsions disappeared upon daily administration of 0.26 mg of pyridoxine.

A *relative* B₆ deficiency, in spite of a normal intake, may result from a variety of causes, among which those discussed below are prominent.

A. High protein diet

The "requirement" for vitamin B₆ is, like that of the vitamins B₁ and B₂, not constant, but depends on other nutritional and metabolic factors (304). Since vitamin B₆ is primarily concerned with amino acid and protein metabolism, it is conceivable that on a protein-rich diet the B₆ requirement is raised and the appearance of the syndrome of deficiency is accelerated and accentuated (49, 91, 378).

B. Thyroxine, 2,4-dinitrophenol, insulin

Abnormal metabolic conditions, such as fever, neoplasms and hyperthyroidism, may also induce a relative B₆ deficiency (271, 324, 517, 554). The treatment of rats with 0.1 to 0.2 mg/kg DL-thyroxine for 10 days resulted in a significant fall in the pyridoxine and Plp content of the liver and heart (322). Apparently, there must have been also an ATP deficiency, since the lowered myocardial and hepatic Plp content could not be raised by pyridoxine administration, but could be raised by ATP. There is a corresponding decrease in the activity of B₆-dependent enzymes, for example cysteic acid decarboxylase, dopa decarboxylase, glutamic-pyruvic transaminase, threonine dehydrase, etc. (95, 232). It appears that the diminished enzyme activity (e.g., dopa decarboxylase activity in the liver) induced by thyroxine results from an impairment of the apoenzyme, too (85, 228).

2,4-Dinitrophenol (DNP), which does not inhibit pyridoxal-phosphokinase *in vitro* (338), is the prototype of an agent which indirectly inhibits the phosphorylation of pyridoxal through uncoupling oxidative phosphorylation with a resultant fall in the synthesis of energy-rich phosphates (ATP). DNP also inhibits the active transport of amino acids (e.g., histidine and methionine) across the intestinal wall. This transport is activated by B₆ derivatives (see section VII). Plp is able to abolish the inhibitory action of DNP. On the other hand, the non-phosphorylated B₆ derivatives—pyridoxal and pyridoxine—are able to prevent inhibition only if they are injected before DPN (246, 247, 502).

Insulin coma is associated with a fall of ATP and creatine phosphate in brain. The diminished decarboxylation of 5-HTP found in brain under conditions of severe insulin hypoglycemia can be attributed to an ATP deficiency which may result in the reduced phosphorylation of pyridoxal (108).

C. Pregnancy

From the third month of pregnancy there is an increased xanthurenic acid excretion, which is particularly large when symptoms of toxemia are present (485, 512, 514, 515). A single dose of 25 mg of pyridoxine can restore the xanthurenic acid excretion to normal for 3 to 5 days. Thiamine, riboflavin, and nicotinic acid are ineffective (82, 512). The chief metabolic product of vitamin B₆, pyridoxic acid, is excreted in normal amounts by pregnant women, but after loading with 25 mg of pyridoxine—"pyridoxine load test" (448)—they excrete significantly less pyridoxic acid than the controls (515). This indicates a retention of pyridoxine as a result of an increased requirement in pregnancy.

Whereas the B₆ content (taken as total activity) of blood, skin, and urine is

normal in pregnancy (160), the Plp content of the plasma and leukocytes is significantly lower than normal (516, 518). Administration of 100 mg of pyridoxine results in a smaller increase in the Plp content of plasma and leukocytes in pregnant women than in nonpregnant women (516).

In this connection it is of interest that estrogens (estradiol disulfate, stilbestrol) are extremely potent inhibitors of vitamin B₆-dependent enzymes. In concentrations as low as 10⁻⁷ M they inhibit dopa decarboxylase, cysteine sulfinic acid decarboxylase, kynurenine transaminase, and phosphorylase "a." Estrone sulfate and the glucuronide of pregnandiol were less potent. According to Mason *et al.* the steroid hormones and Plp compete for the apoenzyme. That the steroid hormones have an affinity for the apoenzyme is also indicated by the fact that they are able to protect the enzyme from proteolytic inactivation by chymotrypsin (325, 326, 327).

D. Vitamin B₆ dependence

Hereditary disturbances in vitamin B₆ metabolism can also lead to a relative B₆ deficiency and its signs. Among these are the cases of epileptiform convulsions which appear as soon as the first hour after birth in certain babies (235, 236). These convulsions are refractory to barbiturates and other antiepileptic therapy but are completely relieved in a few minutes by pyridoxine. A maintenance dose of 2 to 15 mg/day is required to prevent their recurrence (59, 69a, 105, 110, 169, 321, 377, 460, 461). During the convulsions, as in insulin seizures, the cerebral blood flow, oxygen consumption, and respiratory quotient fall, and they return to normal as soon as the convulsions cease after the injection of pyridoxine (480). Tower (499) concluded from these results that the convulsions have their origin in a deficiency of specific substrate. On the other hand, during convulsions due to electroshock or Metrazol brain metabolism is augmented and falls to values below normal only in the postconvulsive phase.

In "absolute" B₆ deficiency (for instance, of dietary origin) associated with convulsions, usually one finds a low blood B₆ content and a decreased urinary vitamin B₆ activity and pyridoxic acid. In the hereditary cases convulsions occur despite a normal B₆ intake and are due to an endogenous B₆ deficiency. After a pyridoxine load Coursin (110) observed, concomitant with a low B₆ blood level, an abnormally high urinary excretion of pyridoxic acid in some cases and of pyridoxal and pyridoxamine in another case. This indicates that the metabolism or excretion of the vitamin is increased above normal in this syndrome, and that this produces an endogenous B₆ deficiency.

These cases probably belong to the group of "inborn errors of metabolism" of which classic examples are phenylketonuria, argininosuccinuria, maple sirup disease, *etc.* (273). This interpretation is supported by observations of the occurrence of several such metabolic anomalies within one and the same family (236, 321, 377, 460).

IV. B₆ DEFICIENCY PRODUCED BY ANTIMETABOLITES

According to Snell and co-workers (335, 519) an ATP-dependent phosphokinase phosphorylates pyridoxine to pyridoxine-5'-phosphate. A flavin-containing enzyme which occurs largely in liver and to a lesser extent in kidney and

brain then oxidizes pyridoxine-5'-phosphate to Plp (519). This enzyme is inhibited *in vitro* by heavy metals (Cu⁺⁺, Hg⁺⁺, *p*-chloromercuribenzoate). Plp itself is a competitive inhibitor. Moreover, 4-deoxypyridoxine-5'-phosphate and pyridoxaloxime-5'-phosphate are inhibitors because they possess a greater affinity for the oxidase than pyridoxine-5'-phosphate (519). More details on the inhibition of phosphokinase are given in section IV A.

A. Structural analogues of vitamin B₆

Without realizing it, Abderhalden (2) was presumably the first to use a B₆ antimetabolite, in 1938. In experiments designed to accomplish the synthesis of vitamin B₁ (thiamine) from its pyrimidine and thiazole moieties *in vivo*, he observed that the injection of the pyrimidine component (2-methyl-4-amino-5-hydroxymethyl-pyrimidine), later named toxopyrimidine (fig. 6, III), produced a typical running fit in mice and rats, *i.e.*, an uncontrollable running movement followed by convulsions. It was only much later that toxopyrimidine was recognized as an antivitamin on account of its structural similarity to the B₆ molecule (317, 318). The toxopyrimidine convulsions can be suppressed by pyridoxine, pyridoxal, or pyridoxamine.

Experimentally, antimetabolites of vitamin B₆ have the advantage that one can produce acute B₆ deficiency syndromes in animals without time-consuming feeding of B₆-deficient diets. Probably the most thoroughly investigated antimetabolite is 4-deoxypyridoxine (fig. 6, I). In this compound the 4-hydroxymethyl group essential for vitamin function (352) is chemically altered. ω -

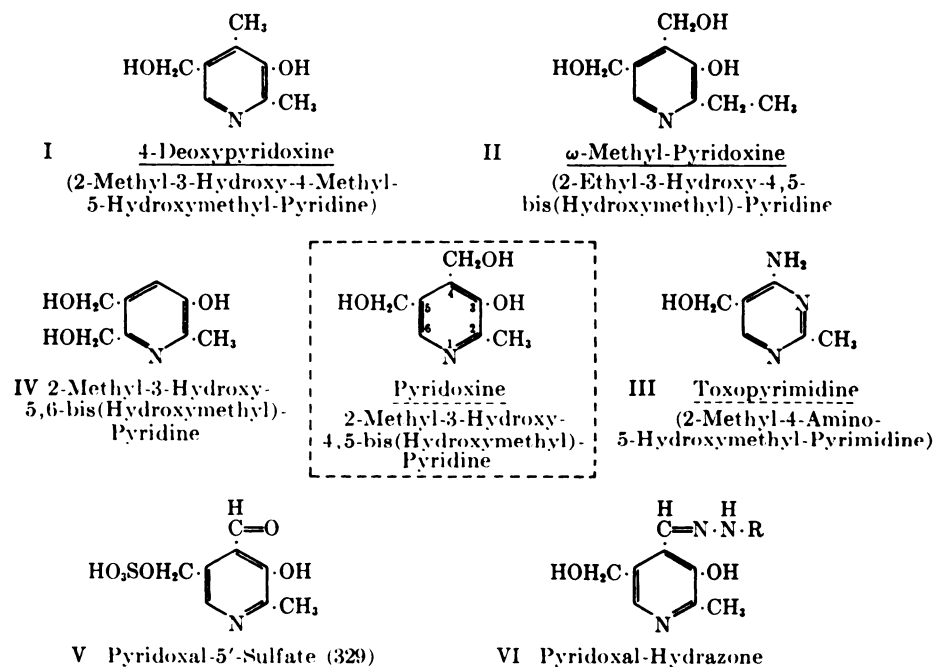


FIG. 6. Vitamin B₆ antimetabolites.

Methyl-pyridoxine (fig. 6, II), however, appears to act as an antivitamin in the rat, causing convulsions, in spite of a normal chemical structure at position 4 (445). On the other hand, this compound can act as a coenzyme of bacterial tyrosine decarboxylase (382).

In animals 4-deoxypyridoxine is highly toxic, especially under conditions of dietary B₆ deficiency (386). Thus the acrodynia in rats is accentuated (143). The lethal action of the antimetabolite on chick embryos can be suppressed by vitamin B₆ (113). Deficiency symptoms after 4-deoxypyridoxine in man reported by Vilter *et al.* (510), consist of dermatitis, glossitis, cheilosis, and neuritis.

The bacterial tyrosine decarboxylase (in *Streptococcus faecalis*) is inhibited *in vitro* only by 4-deoxypyridoxine-5'-phosphate and not by 4-deoxypyridoxine (51, 503). The nonphosphorylated analogue is inactive in concentrations 50,000 times greater than an effective dose of the phosphorylated compound (239). The same is true for toxopyrimidine (318). It is therefore reasonable to assume that 4-deoxypyridoxine acts as an antimetabolite in the animal and human body only after it has been phosphorylated and can compete with the physiologically occurring Plp for the binding sites on the apoenzyme (503). The extent to which a B₆ enzyme is inhibited by an antimetabolite depends upon the relative affinities of the phosphorylated antimetabolite and of Plp for the apoenzyme (474, 475). Therefore, the administration of an antimetabolite *in vivo* does not necessarily inhibit all B₆-dependent enzymes to the same degree, but particularly those whose coenzyme will easily dissociate from the apoenzyme (for review see 116, 238, 475).

Deoxypyridoxine also inhibits the phosphorylation of pyridoxine and pyridoxal by phosphokinase, an enzyme found in high concentration in the cerebral cortex. Yeast, bacterial, and animal phosphokinase can be inhibited by B₆ analogues which themselves undergo phosphorylation (*e.g.*, 4-deoxypyridoxine) and by B₆ analogues that have a high affinity for the phosphokinase although they are not substrates of the enzyme (*e.g.*, 5-deoxypyridoxal) (238, 239, 475). Other analogues (*e.g.*, toxopyrimidine) can be phosphorylated, but are not inhibitors of the phosphokinase (239, 338, 475). Although carbonyl reagents (*e.g.*, hydrazine, hydroxylamine, and semicarbazide) do not inhibit phosphokinase, their oximes or hydrazones (fig. 6, VI), formed in the reaction with pyridoxal *in vitro* (presumably also *in vivo*), are among the strongest inhibitors of the enzyme (336, 337, 338) (see section V D).

B. Substrate analogues

Substrate analogues that have a high affinity to the apoenzyme and are poorly metabolized by the corresponding enzyme may be strong inhibitors by competition with the usual substrate.

In order to elucidate the role of catecholamines in hypertension, many investigations have been performed to block the enzymatic biosynthesis of these amines by inhibition of L-dopa decarboxylase. Thus Clark *et al.* (103, 104, 201) found some compounds chemically related to dopa, *e.g.*, 5-(3,4-dihydroxycinnamoyl)-salicylic acid, caffeic acid, and others, to be rather potent inhibitors of the decarboxylase *in vitro*.

1. *α-Methyl-dopa* (*3,4-dihydroxyphenyl-α-methylalanine*). In 1954, Sourkes (481) investigated dopa analogues which contain a methyl group on the α -C-atom of the side chain. Most active inhibitors are α -methyl-metatyrosine and α -methyl-dopa. The inhibitory action is due to a competition of the inhibitor with the substrate for the apoenzyme. Inactivation of the coenzyme Plp, by formation of Schiff's bases with subsequent transformation to tetrahydroisoquinoline derivatives (see section V A) plays only a minor role. These inhibitors are active also *in vivo*. *Alpha*-methyl-dopa suppresses the increase in blood pressure in response to an intravenous injection of L-dopa (126). It also inhibits the decarboxylation of 5-HTP to serotonin, *in vivo* and *in vitro* (536). The inhibitory action of α -methyl-dopa *in vivo* lasts only a few hours (536). In agreement with this, the decreased serotonin content in brain brought about by α -methyl-dopa returns to normal 6 to 8 hours after injection (471, 472). In contrast, the simultaneously occurring depletion of norepinephrine in brain and heart lasts for days (213, 294, 365, 482).

Although the corresponding amines, α -methyl-dopamine and α -methyl-metatyramine (Aramine), are not inhibitors of dopa decarboxylase, they have the same action as α -methyl-dopa in depleting tissue norepinephrine (403). Therefore the depleting action probably is not due to α -methyl-dopa itself, but to α -methyl-dopamine, produced slowly by decarboxylation of the amino acid (526). Evidence was given by Carlsson and Lindquist (88) that the depletion of the physiological catecholamines as produced by α -methyl-dopa and α -methyl-metatyrosine is brought about by replacement of the catecholamines by the corresponding α -methylated amines, which obviously can also be hydroxylated in the side chain.

Thus, the pharmacological action of α -methyl-dopa, particularly its hypotensive action (178), of which clinical use is made in the treatment of hypertension (21, 86, 177, 302, 469), may be explained by the replacement of the physiological transmitter, norepinephrine, by less vasoactive amines. Controversial results were obtained with α -methyl-dopa in the treatment of patients with carcinoid (69, 132, 470).

In the initial stages of treatment with α -methyl-dopa, fatigue and sedation were encountered (21, 132, 379). Other side-effects can be Parkinsonian symptoms (190). Obviously, also in these cases the methylated amines that originate from α -methyl-dopa cannot substitute functionally for the normal endogenous amines (see section VI).

In hypertensive rats the blood-pressure-lowering effect of α -methyl-dopa was completely prevented by a hydrazine derivative (NSD 1039; N₁-2-hydroxybenzyl-N₁-methyl-hydrazine) (119) which is one of the most potent inhibitors of dopa decarboxylase, presumably by inactivation of Plp (418). This also indicates that α -methyl-dopa itself is not the hypotensive agent. Apparently it has to be decarboxylated to the hypotensive methylated amines. From this it follows that it is not the inhibition of dopa decarboxylase by α -methyl-dopa, but the decarboxylation of α -methyl-dopa by the B₆-dependent dopa decarboxylase which is *de facto* the cause of its hypotensive action.

2. *Phenylketonuria and B₆ enzymes*. Oligophrenia phenylpyruvica, an inborn error in metabolism, is probably due primarily to a lack of the liver enzyme which

hydroxylates phenylalanine to tyrosine. This leads to an increase in the phenylalanine and phenylpyruvic acid concentrations in blood and urine. Clinically this is followed by cerebral manifestations such as oligophrenia and epileptiform convulsions.

In patients with phenylketonuria the values for serotonin in plasma and for 5-hydroxyindoleacetic acid in urine are lowered (392, 393, 394). Also the plasma epinephrine and norepinephrine concentrations (523) as well as the urinary excretion of these amines and of dopamine are decreased (369). The values are restored to normal when these patients are maintained on a diet poor in phenylalanine (60, 275). Feeding high doses of phenylalanine to young rats reduced the serotonin content of brain and blood (23, 522, 563). The explanation of these disturbances in amine metabolism could be an inhibition of amine synthesis. Phenylpyruvate and phenylacetate (but not phenylalanine) inhibit the B_6 -dependent decarboxylation of dopa and 5-HTP (121, 149, 197, 201). In agreement with this, epinephrine synthesis from C^{14} -*p*-tyrosine in adrenal slices is inhibited by phenylpyruvate, while at the same time the accumulation of C^{14} -dopa in the medium indicates dopa decarboxylase inhibition (71). On the other hand, phenylalanine competitively inhibits the uptake of the precursor amino acids for amine synthesis into the cells. Thus, the uptake of tyrosine by brain slices is strongly inhibited by phenylalanine, while that of histidine, arginine, ornithine, and proline is much less affected (343, 372). A single intraperitoneal injection of 1 g of *L*-phenylalanine per kg inhibited the uptake of 5-HTP and thereby the formation of serotonin in the brain to about 60%; *D*-phenylalanine was less potent and phenylpyruvate had no effect (343).

A decrease in cerebral amine content itself appears not to be responsible for the cerebral effects of phenylpyruvic oligophrenia, since rats with an increased serotonin content of the brain induced by treatment with a monoamine oxidase inhibitor (isocarboxazide) showed a reduction of intelligence (high incidence of mistakes in the Hebb-Williams maze test) similar to that of rats kept on a diet rich in phenylalanine and having a low serotonin level in brain (23, 311, 563).

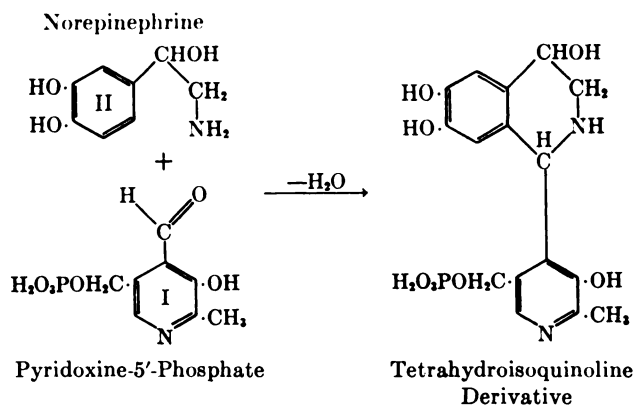
A disturbance in amine metabolism, convulsions, and dermatitis, as observed in patients with phenylketonuria, may indicate a B_6 deficiency. However, long-lasting treatment with vitamin B_6 did not have any therapeutic effect (340).

V. B_6 DEFICIENCY BY INACTIVATION OF PYRIDOXAL-5'-PHOSPHATE (PLP)

The most important functional group of Plp is the aldehyde group in position 4. Blockade of this group inactivates coenzyme function.

A. *Amino acids and amines*

The amount of CO_2 released under anaerobic conditions by brain homogenates is increased if *L*-glutamic acid is added to the medium (see section I B). However, if the homogenate is preincubated for a few minutes with *L*-dopa, dopamine, or norepinephrine, the formation of CO_2 is depressed and is even lower than that seen in the controls to which no glutamic acid has been added (230). This inhibition, which can be reduced by the addition of Plp, is explained by a stoichiometric reaction of catecholamines having a primary NH_2 -group with the



coenzyme, to form Schiff's bases. These rearrange themselves irreversibly to tetrahydroisoquinoline derivatives. Thus the catecholamines as well as the coenzyme lose their pharmacological and biochemical activities (216, 230, 455).

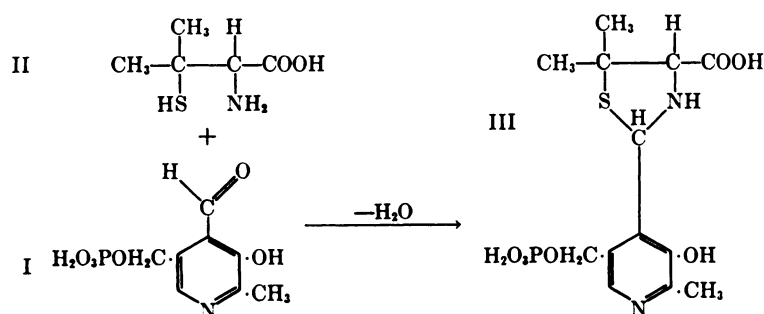
The reaction may be of importance for the *in vitro* estimation of Plp-dependent enzyme activities. In the manometric estimation of dopa decarboxylase, for example, the substrate (dopa) as well as the reaction product (dopamine) is equally able to react with Plp, causing a rapid inactivation of the coenzyme, and thereby to stop the enzymic decarboxylation.

Epinephrine, which is not an inhibitor of dopa decarboxylase or glutamic acid decarboxylase, becomes inhibitory when it is allowed to take up 4 or 6 moles of oxygen per mole. The inhibitory action of oxidized catecholamines cannot be reduced by the addition of pyridoxal-5'-phosphate. Their site of attack is, therefore, probably the apoenzyme (230). This is presumably why Werle and Koch (531) found that epinephrine, catechol, resorcinol, or pyrogallol inhibits histidine decarboxylase and dopa decarboxylase after incubation for 30 minutes in an oxygen atmosphere with tissue enzyme preparations.

B. Penicillamine

D-Penicillamine (α -amino- β -thioisovaleric acid) results from chemical degradation of penicillin and is also a metabolic product of the antibiotic in man (106). It finds clinical application in heavy metal poisoning and Wilson's disease. It accelerates the elimination not only of Cu, but also of Pb, Fe and Hg through chelate formation (18, 154, 451). In rats the L- isomer is significantly more toxic. The oral LD₅₀ for the DL- form is 365 mg/kg, whereas that of the D- isomer is greater than 2500 mg/kg (18, 19). Chronic administration of the L- isomer to rats leads to inhibition of growth, anorexia, and acrodynia as well as to tonic-clonic convulsions (running fit) (19, 549, 550). These effects as well as the accompanying increased xanthurenic acid excretion in the urine can be suppressed by pyridoxine (290, 291).

L-Penicillamine inhibits the absorption of amino acids from the intestine (5, 502) (see section VII). *In vitro* it is an inhibitor of numerous B₆-dependent enzymes, *e.g.*, of transaminases including GOT and GPT (207, 290), of cysteine desulfhydrase (20), and of α -aminolevulinic acid synthetase (303). If the co-

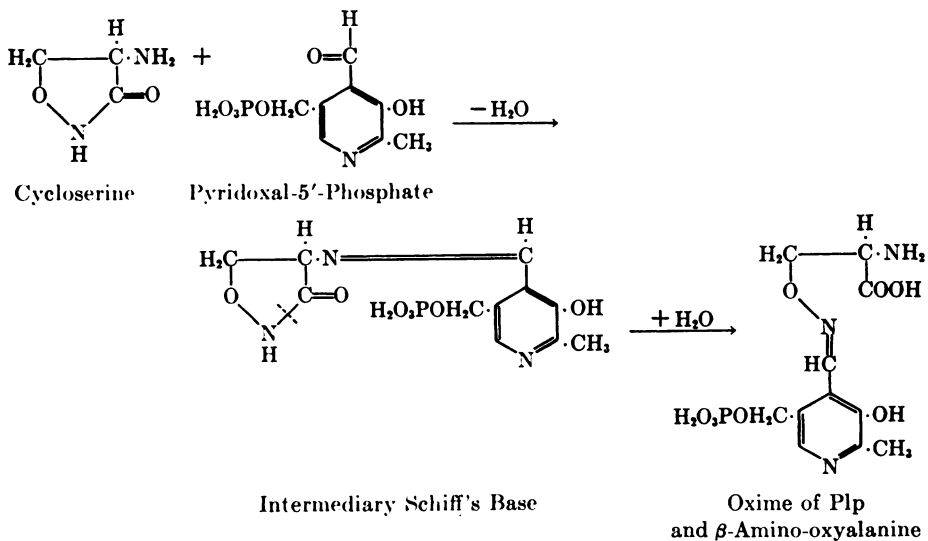


enzyme of these enzymes, Plp, is incubated with penicillamine prior to its addition to the apoenzyme, the enzyme activity is inhibited by both L- and D-isomers. However, if Plp and the apoenzyme are preincubated first, then only L-penicillamine has an inhibitory effect (502). Both isomers (II) are capable of blocking free Plp (I) by forming a thiazolidine derivative (III) nonenzymatically, as is true also for other SH-containing amino acids (316, 320, 551). However, L-penicillamine obviously also inhibits the holoenzyme (502).

Because of the toxicity of L-penicillamine, only the D-isomer should be employed in the therapy of Wilson's disease and heavy metal poisoning (18).

C. Cycloserine

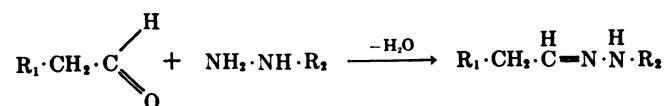
The toxic action of this tuberculostatic drug can be explained at least in part by its inhibition of B_6 -catalyzed enzymatic reactions (144). Among the most important side-effects seen frequently with therapeutic dosage are sedation, psychic disturbances, pareses, and above all epileptic convulsions (157, 161, 308, 407, 568). Cycloserine brings forth seizures and ataxia also in the mouse and rabbit (161). Pyridoxine reduces cycloserine toxicity (144, 433). While Dengler



(125) could not establish an increased xanthurenic acid excretion in patients treated with cycloserine, in rats with high doses there is an enhanced excretion of xanthurenic acid, kynurenine, and 3-hydroxykynurenine due to inhibition of B₆-dependent kynureninase (17). Dopa decarboxylase and glutamic acid decarboxylase are also inhibited *in vitro* by cycloserine (125); the same is true for other decarboxylases and transaminases of animal (380) and bacterial origin (15, 16, 562). Plp abolishes the inhibition. It is probable that, after formation of an unstable Schiff's base and splitting of the isoxazolidinone ring, a stable pyridoxal-oxime of β -amino-oxy- α -alanine results (260, 440a).

D. Hydrazides

Braunstein (76) wrote in 1953: "In our investigations, we made it a rule first to examine the sensitivity towards carbonyl reagents of all enzymatic reactions in which participation of pyridoxal phosphate could be assumed." Hydrazine and hydrazine derivatives of the general structure R—NH—NH₂ are typical carbonyl reagents in that they react with the carbonyl oxygen of aldehydes and ketones to form hydrazones.



However, the hydrazones resulting from the reaction between Plp and hydrazides are still active as coenzymes *in vitro*, as shown by Palm for dopa decarboxylase (391). Whereas on addition of free coenzyme decarboxylation soon comes to an end because of inactivation of Plp by the substrate, the addition of the hydrazone, formed for instance by incubation of isonicotinyldiazide (INH) with Plp, promotes a continuous reaction, which results in quantitative decarboxylation of the substrate. This is presumably due to a slow hydrolysis of the hydrazone, splitting off small amounts of free coenzyme (391). This reaction could be useful in the quantitative manometric estimation of dopa decarboxylase activity. The results were confirmed by Gonnard (182). His assumption that the hydrazone itself is the activating factor (181), however, would be difficult to reconcile with the enzymatic mechanism of decarboxylation (see section I, 1).

Hydrazine itself is of toxicological importance because of its widespread industrial application (131). Mono- and di-substituted hydrazines are of pharmacological and toxicological interest in that a whole series of them have found therapeutic application in the last few years. Among these are the tuberculostatic isonicotinyldiazide (isoniazid, INH) and the monoamine oxidase inhibitor isonicotinyldiisopropylhydrazide (iproniazid, Marsilid) (for review see 400). Compounds such as semicarbazide, thiosemicarbazide, carbohydrazide, and thiocarbohydrazide served for a long time to detect Plp-dependent reactions *in vitro* and *in vivo*. Since most of them penetrate into the central nervous system, they often have been used to examine the role of vitamin B₆ in the CNS. The most characteristic pharmacological action of those hydrazides that penetrate the blood-brain barrier consists in the induction of convulsions (for review see

544). Dieke (129) as well as Parks *et al.* (395) detected the convulsive action of hydrazides such as semicarbazide and thiosemicarbazide, McGrath *et al.* (342) that of hydrazine. Convulsions occurring in tuberculous patients after treatment with INH were reported by numerous authors (94, 112, 286, 292, 440). The hydrazide-induced convulsions have been described by Jenney and Pfeiffer (251) as follows: "These compounds produce maximal, tonic-extensor convulsions which closely resemble the grand mal seizures of epilepsy or supramaximal electroshock and intravenous Metrazol, and differ from most drug-induced seizures, in that an animal may have as many as five maximal seizures before recovery or death." In 1890, Loew (310) had observed in guinea pigs and rabbits the appearance of opisthotonos, intermittent clonic convulsions, and pareses starting in the hind limbs after the injection of hydrazine. He wrote: "The diamide, thus, is like hydroxylamine in numerous organisms a strong poison. Both compounds are chemically characterized by their affinity to aldehydes and ketones."

The convulsive hydrazide doses are very different from species to species. The dog appears to be particularly sensitive, while small rodents are strikingly insensitive. Thus the convulsive doses (mg/kg) of semicarbazide are: dog 10, cat 40, man 40, monkey 60, guinea pig 75. On the other hand, they are 116, 150, and 175 mg/kg for the mouse, rat, and rabbit, respectively (544).

The death of a tuberculous patient who was under treatment with INH, after a single injection of Irgapyrine (phenylbutazone + amidopyrine) led Dienemann and Simon (130) to investigate the influence of INH on the convulsion threshold of amidopyrine, Metrazol, and procaine. In all three cases, INH lowered the seizure threshold. Epileptics appear to be particularly sensitive to the convulsive action of INH (151). Alcohol is supposed to favor the onset of convulsions (189). In animals semicarbazide lowers the threshold for Metrazol but not that for strychnine (251, 419). Also the threshold for electroshock convulsions is lowered after administration of hydrazides (265). Hydrazide convulsions are particularly easy to provoke by acoustical and optical stimuli (32, 265, 413, 419). In combination with flicker light, semicarbazide has been used clinically to produce grand mal seizures in schizophrenic patients resistant to therapy (420).

Like the convulsions in B₆ deficiency, either dietary or hereditary, those brought about experimentally by hydrazides are largely refractory to antiepileptic drugs (barbiturates, hydantoin). Dilantin (diphenylhydantoin) prevented neither semicarbazide nor INH convulsions in rats (251, 419); it only prolonged the survival time. NaBr and Mesantoin (N-methylphenylethylhydantoin) were somewhat active, as was the oxazolidine derivative trimethadione, which has shown itself to be of value in therapy of petit mal epilepsy, and which is one of the most specific antagonists of Metrazol-induced convulsions (148, 183, 251). Hydrazide-induced convulsions have in common with those of B₆ deficiency that they can be suppressed immediately by pyridoxine or its congeners (32, 252). Parks *et al.* (395) were the first to show that pyridoxamine acts as an antidote against poisoning with thiosemicarbazide.

The various hydrazides have very different convulsive potencies (32, 251). The sulfur-containing derivatives, *e.g.*, thiosemicarbazide and thiocarbohydrazide, are much more potent than semicarbazide or carbohydrazide. In mice

after intramuscular injection the ED100's of carbohydrazide (1.66 mmol/kg) and semicarbazide (1.80 mmol/kg) are 10 to 50 times higher than those of the corresponding thio compounds (0.038 and 0.121 mmol/kg) (32). The ED100 of the B₆ antimetabolite 4-deoxypyridoxine is 1 mmol/kg. The convulsive action of hydrazides can be prevented by simultaneous injection of pyridoxine. On a molar basis, a dose of pyridoxine approximately one-half that of semicarbazide, carbohydrazide, thiocarbohydrazide, cyanoacetic hydrazide, or hydrazine is necessary to reduce the ED100 to the ED50. On the other hand for thiosemicarbazide, deoxypyridoxine, and isopropylhydrazine, the ED100 can be correspondingly reduced by pyridoxine in 1/8th, 1/22nd and 1/42nd of the molar dose of the hydrazine (32).

INH occupies an exceptional position in that even a molar dose of pyridoxine two times that of INH does not prevent INH convulsions in the rat and mouse (32, 134, 319, 413). On the other hand some authors showed that pyridoxine antagonized the convulsant effect of INH in the dog (545) and decreased its toxicity in mice and man (419). Plp or pyridoxal actually accelerates the appearance of INH seizures and produces convulsions in animals given subconvulsive doses of INH (see section V D, 2) (32, 107, 375).

1. *Hydrazides and GABA metabolism.* Since Plp is also the coenzyme of glutamic acid decarboxylase, it could be expected that hydrazides would affect cerebral glutamic acid metabolism, particularly the formation of GABA. According to Killam (262, 264), convulsive doses of hydrazide lower GABA content of the brain through inhibition of glutamic acid decarboxylase. This has been confirmed for a variety of hydrazides (32, 43, 139, 150, 263, 333, 375, 427, 428, 544). Roberts *et al.* (432) could show an inhibition of the decarboxylation of intracisternally injected C¹⁴-L-glutamic acid to GABA after thiosemicarbazide in rats. This suggested that hydrazide-induced seizures were caused by a diminished cerebral GABA content. However, other investigators showed that the causal relationship between glutamate-GABA metabolism in brain and the occurrence of convulsions is more complicated. In mice, 1) the GABA content of the brain remains unchanged after convulsive doses of thiosemicarbazide; 2) the simultaneous injection of thiosemicarbazide and a convulsion-preventing dose of pyridoxine significantly lowers the GABA content of the brain; and 3) the GABA content of the brain, lowered by convulsive doses of other hydrazides, is still more lowered by the simultaneous injection of a convulsion-preventing dose of pyridoxine (32). Obviously, pyridoxine in protective doses activates not only the decarboxylation of glutamic acid, with a formation of GABA and an increase in the GABA content of the brain, but also, and to a greater extent, the transamination between GABA and α -ketoglutaric acid, leading to the disappearance of GABA and thus to a decrease in the GABA content of the brain (see fig. 2). Thus, it appears that it is not the cerebral GABA content as such that is of decisive importance for the susceptibility of cerebral motor centers to convulsive stimuli but the rate of the turnover in the glutamate-GABA shunt (32).

Confirmation of these findings is given by experiments of Rindi and Ferrari (425), in which pyridoxamine could prevent toxopyrimidine-induced convulsions but not the accompanying fall in the GABA content. Moreover, experiments of

Baxter and Roberts (45), as well as Maynert and Kaji (333), showed that pyridoxine or pyridoxal prevented the convulsions induced by semicarbazide and thiosemicarbazide, but not the fall in cerebral GABA.

The lack of a causal relationship between the cerebral GABA content and seizure threshold is also supported by the fact that mouse strains susceptible to spontaneous seizures had approximately a 50% higher cerebral GABA content than other nonsusceptible strains (371). On the other hand, in susceptible strains the Plp content decreased more rapidly in dietary B₆ deficiency (314). Also, in cats there has been found no correlation between GABA content and electroshock thresholds in various cortical areas. The hippocampus, which has a particularly low seizure threshold (convulsions can be induced purely through mechanical excitation), has an especially high GABA content (46).

That the activity of GAD alone does not determine the appearance of convulsions, is shown in that various hydrazide-induced convulsions appear at completely different levels of GAD activity, that is, at different degrees of inhibition. After equipotent convulsive doses, the enzyme activity of mouse brain was inhibited only by 23 and 43% with thiocarbohydrazide or thiosemicarbazide, respectively, while after carbohydrazide, semicarbazide, or INH, it was inhibited by 65 to 70% (32). From experiments by Killam (262) with thiosemicarbazide in the cat, on the other hand, it may be concluded that there exists a parallel between the degree of GAD inhibition and the appearance of electroencephalographic seizure potentials. Incubation of the hydrazides with Plp showed, in agreement with the affinity constants reported by Wiegand (538), that semicarbazide, for instance, has a greater affinity to the coenzyme than the sulfur-containing compound, although the latter possessed a much higher convulsive potency (32). This is perhaps explained by the greater lipid solubility of the thio- derivatives.

The crucial experiment to support the hypothesis that a depression of the GABA turnover rather than a lowering in the GABA content of the brain is the cause of hydrazide convulsions, would be to inhibit GABA-T specifically or at least to a higher degree than GAD, thereby increasing the cerebral GABA content, and still produce convulsions. Hydroxylamine *in vivo* inhibits the transaminase more strongly than the decarboxylase. GABA-T inhibited *in vitro* by hydroxylamine cannot be reactivated as readily as GAD. This probably results from the higher affinity of Plp to the transaminase apoenzyme; therefore the dissociation of inactivated coenzyme (pyridoxal-phosphate-oxime) from the apoenzyme, and its replacement by Plp would be slower than in the case of GAD (44, 45). In rats, the administration of hydroxylamine results in convulsions shortly after the injection, while 1½ hours later the cerebral GABA content is raised to 180% of normal (41, 42, 137). This action is extraordinarily species- and strain-dependent. In Wistar rats, the rise in GABA amounts to only 38%, in Sprague-Dawley rats to 9%; in Royal-Hart rats and mice there is no increase at all (162, 494). The increased GABA content and the inhibition of GABA-T caused by hydroxylamine is associated with an increase in the seizure threshold for electroshock and Metrazol, but not for thiosemicarbazide (42, 137). This hydrazide led to convulsions in rats, even if the GABA content remained normal

or was raised by the simultaneous injection of hydroxylamine (42). Similar results were obtained in mice with diacetylmonoxime; thiosemicarbazide convulsions could not be prevented despite a normal or even augmented GABA content of the brain (162). A dose of 100 mg of hydrazine hydrochloride per kg raised the GABA content of mouse brain by 300% after 12 hours, but did not protect the animals against Metrazol- and semicarbazide-provoked convulsions (333).

A very active inhibitor of GABA-T is β -amino-oxyacetic acid ($\text{NH}_2\text{OCH}_2\text{—COOH}$), which causes vomiting, sedation, and muscular paralysis in dogs and cats at low doses, while higher doses result in convulsions which are correlated with a strong elevation of cerebral GABA levels (44, 521). DaVanzo *et al.* (118) found that in mice, rats, and cats, amino-oxyacetic acid exerts a certain protective action against methionine-sulfoximine and semicarbazide convulsions, which is maximal 1 hour after the administration of amino-oxyacetic acid. After 8 hours, however (when the GABA levels were maximally increased), this protective effect had disappeared.

According to Massieu *et al.* (328), another hydrazide, L-glutamic acid γ -hydrazide, inhibits GAD more strongly than GABA-T, *in vitro*. However, it elicits convulsions, although the cerebral GABA content increases to 6 to 10 times the normal. Also α, γ -diaminobutyric acid, which inhibits GAD but not GABA-T *in vitro*, led to fatal convulsions with an increase of brain GABA by 200 to 300% after a latency of 24 hours. It has been suggested that diaminobutyric acid is transformed in the body to an inhibitor of GABA-T (258, 344).

It appears that there is a relationship between hydrazide-induced seizures and a lack of cerebral Plp, the coenzyme of GAD and GABA-T. Hydrazides inactivate the coenzyme by hydrazone formation and impair its synthesis by inhibition of phosphokinase, the enzyme necessary for the phosphorylation of pyridoxine and pyridoxal. As found by Snell and co-workers (336, 337, 338), hydrazones as well as oximes of pyridoxal are potent inhibitors of phosphokinase. Indeed, after the administration of hydrazides the corresponding hydrazones of pyridoxal and Plp have been detected in brain and urine (28, 543, 544). In agreement with this, Bain and Williams (28, 544) found that after INH, deoxy-pyridoxine, or toxopyrimidine the cerebral Plp content was diminished whereas the pyridoxal and pyridoxine concentrations were increased, possibly due to an inhibition of phosphokinase. Decreased coenzyme levels in brain were found also if seizures were suppressed by high doses of phenobarbital (28). Also the convulsant action of structural analogues of vitamin B₆, such as 4-deoxypyridoxine, toxopyrimidine, and others is due to a competitive displacement of Plp from the apoenzyme as well as to an inhibition of phosphokinase and possibly of pyridoxine oxidase (338, 519).

It seems reasonably safe to conclude from what has been stated that B₆ deficiency can be the cause of convulsions. However, the causal connection between the occurrence of seizures and alterations in the metabolism of GABA remains obscure. Although a decrease in brain GABA may result in a lack of a cerebral inhibitor, thereby lowering the seizure threshold, the GABA content of the brain *per se* is not the decisive factor, because convulsions have been found to occur with decreased as well as with increased concentration of brain GABA.

Therefore, it appears doubtful that the role of GABA as a central synaptic inhibitor is of importance for the excitability of motor centers.

The main point could be that a decrease of the GABA content by inhibition of GAD as well as an increase of the GABA concentration by inhibition of GABA-T reduces the turnover of the brain specific glutamate-GABA shunt, which participates to about 40% in the oxidative metabolism of the brain. From this it follows that under all circumstances B₆ deficiency will lead to a diminished oxygen consumption in the brain. This has indeed been observed in kittens with a dietary B₆ deficiency (345) as well as in brain slices of rats pretreated with hydrazides. In both instances, the GABA content was decreased. On the other hand, the administration of α,γ -diaminobutyric acid to rats, causing an increase of the cerebral GABA concentration, nevertheless produced a diminished oxygen consumption (344). It has previously been mentioned that a decreased oxygen consumption of the brain was observed in a "B₆-dependent" patient (480). Seizures due to B₆ deficiency can be blocked only by vitamin B₆, but not by the administration of GABA (258, 408) or glucose (251). This again proves that the decreased oxygen consumption and the seizures are not caused by a lack of substrate but rather by a deficiency of the enzymatic catalyst of the brain turnover in the glutamate-GABA metabolism.

2. *INH convulsions.* The convulsive action of INH in animals was first described by Benson *et al.* (54) and by Rubin *et al.* (441). INH is frequently used in suicide attempts: 200 mg/kg are said to be the absolute convulsive dose which leads eventually to death by respiratory paralysis (288, 292). While INH convulsions, as already mentioned, cannot be suppressed by pyridoxine, compounds with a carbonyl group, *e.g.*, pyruvic acid, are reported to exert a protective action, just as α -ketoglutarate and pyruvate antagonize semicarbazide convulsions (39, 251, 465). This protection is probably due to the formation of hydrazones, occurring already outside the brain. The hydrazone formed from INH and pyruvate has been detected as a urinary excretion product (244).

The failure of pyridoxine or pyridoxal to protect against INH convulsions in the experiments of Quadbeck and Sartori (413) could be caused by a too low dosage of Plp, since pyridoxal was given only in one-fourth of the convulsive INH dose. A part of the circulating Plp may be inactivated by INH with hydrazone formation already in the peripheral blood, so that too little free coenzyme reaches the brain to prevent the convulsive action of INH. However, even when a dose (750 mg/kg) is injected which is twice the convulsive INH dose (200 mg/kg) on a molar base, convulsions occur in mice. Indeed, they occur earlier than after INH administration alone. Subconvulsive doses of INH together with Plp cause seizures in all animals (32). Plp appears to accelerate the penetration of INH into the brain. The convulsion-promoting action of pyridoxal phosphate occurs only with INH—not, for example, with semicarbazide (32). On the other hand Dixon and Williams (130a) have shown that the hydrazones of pyridoxal and Plp with INH and other hydrazides are 10 to 100 times more potent as convulsants than the parent hydrazides upon intracerebral injection. However, pyridoxal and Plp are by themselves convulsant upon intracerebral injection (32, 285).

INH, acting as a hydrazine, can inhibit B₆-dependent reactions, and, as a pyridine can also inhibit those catalyzed by NAD (diphosphopyridine nucleotide) (256). Thus, its convulsive action could be caused by an interference with the energy-delivering reactions of glycolysis and of the Krebs cycle, which are necessary for the synthesis of energy-rich phosphate compounds. It is known that anoxemia and glucose deprivation can lead to convulsions. In favor of this assumption is the fact that the simultaneous injection of 400 mg/kg nicotinamide (two times the molar convulsive dose of INH) protects 50 % of the animals from the convulsive action of INH (ED₁₀₀ = 200 mg/kg) (32). Indeed, the INH analogue of NAD has been isolated from liver and brain homogenates after incubation with INH (256). It is of interest that C¹⁴-labeled INH accumulated in the hippocampus of cat brains (38), and that another analogue of nicotinamide, 3-acetylpyridine, which also produces convulsions and can be incorporated by nucleotidase into the NAD molecule, produced histological lesions in the region of the hippocampus and hypothalamus (82a, 256).

3. *INH neuritis*. Although B₆ cannot protect against the central effects of INH intoxication (convulsions, psychoses, ataxia, and somnolence) it has a protective action against INH-induced peripheral neuropathies (440). The toxic neuropathy resulting from the therapeutic application of INH in tuberculosis is characterized by sensory disturbances and pareses (for review see 292). According to Biehl and Vilter (61) the frequency with which the neuritic symptoms appear is dose-dependent: 10 % with 6 to 10 mg/kg daily; 20 % with 15 mg/kg; 40 % with more than 20 mg/kg daily. Histopathologically, the neurotoxic action of INH manifests itself in a primary degeneration of the myelin sheath with myelin fragmentation and formation of droplets of neutral fat. Inflammatory reactions are not seen. The distal portions of the spinal nerves are most strongly affected, especially in the region of the sciatic plexus. Later the spinal ganglia and cranial nerves are also damaged (268, 269, 270, 289, 564, 565, 566). Similar injury can be produced by the local application of INH to the exposed sciatic plexus of the rat (268).

While some authors believe that high doses of nicotinamide have a curative effect in INH neuritis, others deny any therapeutic effect (for review see 26). In experimentally induced INH neuritis in the rat nicotinamide had only a slight protective action (566). On the other hand, there exist numerous clinical and experimental observations which indicate a causal relationship between INH neuritis and B₆ deficiency. Vilter *et al.* (61, 510) concluded from the similarity between the neuritic effects resulting from high doses of deoxypyridoxine (hot feet, electric foot syndrome, hyporeflexia, weakened vibration perception, *etc.*) and those produced by INH that both were caused by B₆ deficiency. In monkeys dietary B₆ deficiency or deoxypyridoxine leads to swelling of the Betz cells of the motor cortex, and to a lesser degree of the small pyramidal cells, and to eccentric displacement of the nucleus and loss of Nissl substance (508). In peripheral nerves, there is degeneration of the myelin sheath. After deoxypyridoxine, the axon cylinder is also affected, but basal ganglia, cerebellum, and spinal cord are not. Klinghardt observed marked loss of medullary substance in

the exposed sciatic plexus of the rat after local application of a deoxyripyridoxine depot (268).

INH-treated patients excreted more B₆ than normal, probably in the form of the hydrazone. Pyridoxic acid excretion was unaltered. Following tryptophan loading, more xanthurenic acid was sometimes excreted. Medication with pyridoxine resulted in the disappearance of the neuritis and of the increased xanthurenic acid excretion (61, 158, 292, 404). B₆ derivatives had a protective effect against INH neuritis also in rats (564, 565, 566).

In many tuberculous patients, the urinary output of xanthurenic acid was already increased before the INH treatment (26). An increased protein catabolism in tuberculosis appears to be the basic disturbance which in turn leads to a latent B₆ deficiency and in this way becomes the pacemaker of the toxic action of INH (26). Although numerous investigations indicate a B₆ deficiency as the cause of INH neuritis, the absence of skin lesions (acrodynia, *etc.*) speaks against this hypothesis. Klinghardt (268, 269) has discussed the possibility of a disturbance in lipid metabolism through coupling of the hydrazine group of INH with fatty acid aldehydes, particularly those of plasmalogens. Although it is known that Plp is the coenzyme for the synthesis of sphingosine, a constituent of sphingomyelin, it has to our knowledge not been considered that a disturbance of this enzymatic reaction through INH could be a pathogenic factor in INH neuritis (see section I, F).

4. *Isopropylhydrazine convulsions.* As already mentioned, the convulsive dose (ED100) of isopropylhydrazine (IPH) is particularly high (3.64 mmol/kg intramuscularly). The reason could be that this hydrazine—probably the active group of the monoamine oxidase (MAO) inhibitor iproniazid—is by itself a potent inhibitor of this enzyme (399). Blockade of brain MAO, however, increases brain amines (dopamine, norepinephrine, serotonin) and hence presumably elevates seizure threshold. Thus, by elevating the amine content in brain, IPH might antagonize to a certain degree its own convulsive action. In favor of this assumption is the fact that after pretreatment of mice with iproniazid, which is a MAO inhibitor only, L-dopa suppressed the tonic phase of hydrazide-induced seizures (32).

5. *Hydrazides and amine metabolism.* Whereas hydrazides, potent inhibitors of amino acid decarboxylases *in vitro*, lower the cerebral GABA content by inhibition of GAD *in vivo*, it appears that decarboxylation of aromatic amino acids, *e.g.*, dopa and 5-HTP, is little influenced as indicated by a normal dopamine and serotonin content of the brain even after convulsive doses of hydrazides. Thus, in mice the serotonin content of the brain remains nearly unaltered after administration of 250 mg of semicarbazide per kg (525) a dose which definitely lowers the cerebral GABA content. The same is true for convulsive doses of INH (399). Also dietary B₆ deficiency which diminishes GAD activity and hence the GABA content of brain (345, 428) leaves decarboxylation of aromatic amino acids (L-dopa, 5-HTP) unimpaired so that normal dopamine and norepinephrine levels in brain are maintained (483, 535). In B₆-deficient chickens with a normal cerebral dopamine content, however, the brain serotonin is decreased, a finding that indicates a higher turnover of this amine (525).

Even the most potent decarboxylase inhibitors, such as the recently developed hydrazine and hydroxylamine derivatives, *e.g.*, NSD 1024, 1034, and 1035, even in high doses, lowered the endogenous amine content of the brain only a little (79, 84). Obviously the remaining decarboxylase activity is sufficient to maintain the very low normal level of amines.

After phenelzine (phenylethylhydrazine), which inhibits the decarboxylation of injected 5-HTP in mice to 50%, the serotonin content of the brain is even increased, since this hydrazine inhibits MAO more strongly than 5-HTP decarboxylation (133). In hydrazides with the hydrazine group substituted, *e.g.*, iproniazid or nialamide, the inhibitory activity on B₆-dependent enzymes is still more reduced as compared to their strong MAO-inhibiting action (for review see Pletscher, 400).

VI. VITAMIN B₆ AND THE PARKINSON SYNDROME

Relationships between vitamin B₆ and the Parkinson syndrome in man result from the fact that practically the total cerebral dopamine content, formed by B₆-dependent decarboxylation of dopa, is localized in the caudate nucleus, putamen and globus pallidum (see section IA, 1) (58, 87, 446). The suspicion that dopamine is of importance for the function of the extrapyramidal system is supported by the finding that the dopamine content of the caudate nucleus and putamen was strongly reduced in Parkinsonian patients at autopsy (57, 136, 233). Parkinsonians also excrete less dopamine and 5-hydroxyindoleacetic acid in the urine than do normal controls (34, 35, 36). Thus, it is understandable that all measures which tend to increase the dopamine content of the brain have been shown to be therapeutically active, in that they relieve, at least temporarily, the cardinal symptoms of the disease: rigor, tremor, and akinesia. Particularly useful is the administration of dopa, which is decarboxylated in brain and causes an increase in cerebral dopamine. Simultaneous treatment with B₆ and monoamine oxidase inhibitors may even produce a stronger elevation of the endogenous dopamine levels in the brain (37, 50, 57, 64, 200). Thus the observation in 1929 by Behringer and Wilmanns (50) of a therapeutic effect of harmine can perhaps be explained by the fact, as we know today, that harmine is a reversible inhibitor of monoamine oxidase (for review see 400).

Vitamin B₆ (50 to 100 mg/day) was first applied to the treatment of Parkinsonism with success in the 1940's (253). It appears that better remissions have been obtained with higher doses (600 to 1400 mg/day) (29, 152, 188, 200, 442). Recently, Hartmann-von Monakow (200) has come to the conclusion that vitamin B₆ is indicated in all cases of Parkinsonism not resulting from arteriosclerosis. This therapy is particularly useful in the postoperative treatment of patients in whom the globus pallidum has been stereotaxically electrocoagulated. In these cases, the B₆ medication acts to resolve the walking and speech disturbances and the loss of motivation and memory that occur after the operation. The fact that high doses of reserpine (cerebral amine depletion) may produce Parkinsonian symptoms (*e.g.*, akinesia, rigor, and catalepsy) which are relieved by the administration of dopa and monoamine oxidase inhibitors in animals

and man (89, 124, 341), also points to a causal relationship between cerebral dopamine and extrapyramidal function.

Against Parkinsonian symptoms brought about by administration of phenothiazine derivatives, dopa plus B₆ was only slightly active or almost inactive (341). Unlike reserpine, the phenothiazines do not produce amine depletion, but are assumed to block not only peripheral but also central adrenergic effects. Phenothiazine derivatives which contain a piperazine moiety (*e.g.*, trifluoperazine, prochlorperazine, perphenazine, thiopropazate and thioproperazine) and (unlike chlorpromazine, promethazine, and others) have only a weak antihistaminic action, most frequently cause such extrapyramidal side effects. In this connection, it is of interest that bulbocapnine, which induces a catatonic state in man and animals, raises the histamine content of the CNS 4-fold in the albino rat (93), presumably by inhibition of histaminase through interaction with its coenzyme Plp (92). The catatonic state as well as the extrapyramidal effects of phenothiazines with weak antihistaminic action, as mentioned above, can be hindered or abolished by strong antihistaminic agents such as diphenhydramine (341).

These results could indicate that the antagonism between adrenergic and histaminergic actions, long recognized in the periphery, as for example on smooth muscle, is perhaps of importance in definite functions of the CNS. In agreement with this would be the fact that anticholinergic drugs such as atropine, scopolamine, Parpanite (caramiphen) and Artane (trihexphenidyl·HCl), have been widely used with success in the treatment of human Parkinsonism for a long time. An absolute dopamine deficiency in the key areas of the extrapyramidal system (reserpine, human Parkinsonism) or a "functional" dopamine deficiency (phenothiazine derivatives) appears to result in an preponderance of cholinergic and histaminergic mechanisms over adrenergic ones leading to the appearance of Parkinsonian symptoms. Recent investigations indicate a further possibility. Like bulbocapnine, papaverine and other opium alkaloids of the isoquinoline type (laudanone, laudanone), as well as mescaline, can produce a hypokinetic rigidity (145). These alkaloids have in common one, two or—in the case of mescaline—three methoxy groups. According to Ernst (145), the subcutaneous injection of 4-methoxytyramine or 3,4-dimethoxydopamine provokes a striking syndrome of hypokinetic rigidity and a catatonic state in cats, sometimes preceded by tremors. Therefore a disturbance of the methyltransferase system, discovered by Axelrod, resulting in an O-methylation of dopamine in the *para* position with or without O-methylation in the *meta* position, might create such a hypokinetic rigidity also in human Parkinsonism.

VII. VITAMIN B₆ AND ACTIVE TRANSPORT OF AMINO ACIDS

A great number of experiments indicates that vitamin B₆ not only is of importance as a coenzyme for the intermediary metabolism of amino acids, their synthesis and degradation, but also is necessary for the intracellular uptake and accumulation of amino acids by an active, energy-requiring transport (99, 210, 540). The well-known fact that tissue extracts can have a 5- to 10-fold higher

amino acid concentration than the blood plasma indicates the possibility of amino acid uptake against a concentration gradient (505a).

Vitamin B₆ also appears to be necessary for a normal absorption of amino acids by the intestine (99). Thus, less L-alanine was absorbed from the isolated, perfused intestine of guinea pigs pretreated with deoxypyridoxine, as compared with the intestine of control animals (159). Jacobs *et al.* (246, 247, 248) found that pyridoxine increased the absorption of methionine in normal rats (intestine perfused *in situ*), and restored to normal the diminished absorption in B₆-deficient animals (deoxypyridoxine or dietary). L-Penicillamine inhibited the absorption of histidine; again pyridoxine was able to abolish this inhibition (5, 502). In B₆ deficiency also the incorporation of amino acids into protein is inhibited. For instance, enrichment of radioactive phenylalanine in the tissues after injection of C¹⁴-phenylalanine into B₆-deficient rats was substantially less than in controls (361). That the underlying mechanism was an inhibition of the incorporation of the amino acid into tissue protein is indicated by the fact that after longer experimental periods the specific activity incorporated into protein remained far below normal, despite a high radioactivity in the protein-free supernatant (361).

A "model amino acid"— α -aminoisobutyric acid—which does not naturally occur in the organism and is not metabolized, but is governed by the same laws of active transport as the natural amino acids, is particularly suitable for such experiments. Thus, if one injects C¹⁴- α -aminoisobutyric acid (AIB) into B₆-deficient rats, the serum level of AIB is about 40% higher than in the controls over a long period of time, while the concentration in heart, skeletal muscle, intestine, and kidney is significantly lower (376, 424).

A "membrane" (cell barrier) of particular pharmacological interest is the blood-brain barrier. B₆ appears to be of importance for its permeability too. *Gamma*-aminobutyric acid (GABA) does not penetrate from the blood into the brain in normal adult animals (171, 468, 500). The GABA content of the brain remained unchanged after i.v. injection of 200 mg/kg GABA to rats, but increased almost 200% if the animals were pretreated with the B₆ antimetabolite 4-methoxymethyl-pyridoxine (165). It is an open question whether this increase in the permeability of the blood-brain barrier is due to a damage to the cell structures or to the convulsions induced by the antimetabolite (408).

Experiments on Ehrlich ascites tumor cells *in vitro* have elucidated the mechanism by which vitamin B₆ influences the active transport of amino acids. One of the most important observations was that pyridoxal or Plp, but not pyridoxine or pyridoxamine, stimulates the uptake of glycine and α , γ -diaminobutyric acid by ascites tumor cells (423). Such cells from B₆-deficient animals were clearly less able to accumulate glycine. The uptake could be restored to normal by the addition of pyridoxal. Deoxypyridoxine inhibited the uptake (101).

Oxender and Christensen (387) found a stimulating action of pyridoxal on amino acid transport across a model of an epithelial membrane, a layer of ascites cells, 4 to 5 cells thick, sucked onto a millipore filter. If pyridoxal was added to one side of the membrane after the equilibration of glycine on both sides, there occurred an enrichment of the glycine concentration on the other side. The reac-

tion underlying this stimulatory action of pyridoxal is possibly the same by which it catalyzes enzymatic reactions, namely the formation of Schiff's bases between amino acids and pyridoxal. It is known that pyridoxal can form chelates with cations like Na^+ , K^+ , Li^+ , Ca^{++} , Mg^{++} , or Mn^{++} . Since the stimulated uptake of amino acids produced by pyridoxal is accompanied by an enhanced potassium efflux, the hypothesis was put forward that there is a combined transport of all three components in the form of a chelate (98, 100, 389).

The fact that carbonyl reagents (hydroxylamine, semicarbazide) do not influence the stimulating action of pyridoxal on the cellular uptake of amino acids appears to be an argument against the formation of a Schiff's base as the primary reaction. Difficult to reconcile with a carrier function of pyridoxal is the fact that the uptake of pyridoxal into cells occurs rapidly, but is soon arrested, while its stimulatory action on amino acid uptake continues for a long time thereafter (387, 388, 390). The amino acid influx is therefore independent of the pyridoxal influx. P1p stimulates amino acid transport but weakly. It is taken up by the cell only in very small amounts, because it is probably fixed on the cell surface (390). Hence, it has been discussed recently whether the action of pyridoxal in promoting cellular accumulation of amino acids could be the result of an inhibition of amino acid efflux (210, 211, 217, 218, 388).

VIII. VITAMIN B₆ AND TUMOR GROWTH

Since B₆ in its phosphorylated form is the coenzyme in numerous enzymatic reactions concerned with the synthesis and degradation of amino acids and proteins, the prerequisite for growth, it was to be expected that this vitamin would be of importance for tissue growth and that a deficiency would lead to inhibition of growth, benign as well as malignant.

Dietary B₆ deficiency inhibits the growth of chemically induced as well as implanted tumors. Treatment with *p*-dimethylaminoazobenzene leads to the formation of hepatomas in normal rats but not in animals with a B₆ deficiency (359). With normal intake of B₆, the expansion of the liver tumors was maximal (559). The carcinogenic action of methyleholanthrene in rabbits was also diminished in B₆ deficiency (267). Implants of sarcoma 180 (65), as well as Flexner-Jobling carcinoma, Yale adenocarcinoma, or mouse fibrosarcoma did not take in B₆-deficient rats (267). Growth of "sarcoma 180 solid" in mice was inhibited only when the animals were placed on a B₆-deficient diet 2 weeks before implantation of the tumor (355). In the controls, there was rapid tumor growth which led to death in 5 weeks. In the B₆-deficient animals, there was an initial growth of the implant which ceased after 2 weeks. The tumor was completely reabsorbed and disappeared. All B₆-deficient animals survived. If they were then placed on a B₆-containing diet the tumors did not reappear.

That B₆-deficient diets inhibit the growth of "sarcoma 180 solid" is of particular interest because this tumor is resistant to most antimetabolites such as antivitamin, and folic acid, pyrimidine, or amino acid antimetabolites, with the exception of 6-mercaptopurine (357). On the other hand, B₆-deficient diets had no effect on the survival time of experimental animals with various ascites tumors, adenocarcinoma 755, and Ehrlich carcinoma (solid), although a clear inhibition

of growth did occur. In Ridgway osteogenic sarcoma and Walker carcinosarcoma, B₆-deficient diets had no detectable effect (357). It is possible to develop a subline of sarcoma 180 solid by passing the tumor 20 times through B₆-deficient animals; it then shows unrestrained growth in B₆-deficient animals. When both the normal and subline tumors are implanted in B₆-deficient animals, the subline tumor takes, but the normal tumor does not (356).

B₆ antimetabolites are also able to inhibit tumor growth. Weir and Morningstar (524) reported marked remission of lymphatic leukemia in a patient after deoxyripyridoxine treatment. However, Gellhorn and Jones (172) observed no therapeutic effect with deoxyripyridoxine and B₆-deficient diets in patients with lymphosarcoma or acute leukemia. In animal experiments the therapeutic action of B₆ deficiency can be shown (357).

In rats the Murphy lymphosarcoma was completely reabsorbed after deoxyripyridoxine or methoxyripyridoxine treatment (489). Deoxyripyridoxine suppressed the ability of fibrosarcoma transplants to take without producing evidences of B₆ deficiency (309). Hydrazides, such as INH (155), can also inhibit tumor growth. Deoxyripyridoxine potentiates the INH action (167). Mouse ascites tumor cells are damaged as soon as 9 hours after a single intraperitoneal injection of INH (466). Also hydrazones, such as pyridoxal-methyl- and pyridoxal-dimethylhydrazone, act as cytostatic agents against sarcoma 180. The isonicotinylhydrazone of pyridoxal was also bacteriostatic against various anaerobes. The hydrazones of pyridoxal and semicarbazide, as well as those of Plp with INH and thiosemicarbazide, proved to be inactive. The same was true of the dimethylhydrazone of Plp (541, 542). Combined therapy (dietary deficiency, B₆ antimetabolites, hydrazides) can have stronger therapeutic effects as compared with the action of the single agents. This was shown by Brockman *et al.* (78) with sarcoma 180. The experiment involved the simultaneous administration of deoxyripyridoxine and a hydrazide, *e.g.*, benzoylhydrazide, *p*-aminosalicylic acid hydrazide or cyanoaceto-hydrazide, together with a B₆-deficient diet. The strongest inhibition was seen with the combination of B₆-deficient diet, hydrazide and deoxyripyridoxine-5'-phosphate. The phosphorylated deoxyripyridoxine plus hydrazide was more active in animals on a normal diet than the nonphosphorylated compound.

In the adenocarcinoma of the mouse the B₆ content of the tumor was lower (0.66 $\mu\text{g/g}$) than that of the organ with the lowest value, *i.e.*, lung (0.73), spleen (0.88), liver (5.2 $\mu\text{g/g}$) (323, 463). Similar results were obtained in human tumors regardless of their site of origin (30). The B₆ content of liver mitochondria was reduced to one-fifth of normal after feeding 4-dimethylaminoazobenzene (406). However, the NAD and NADP content of tumor tissue was also abnormally low (250). Thus, various enzyme activities are diminished in the tumor tissue. The activity of the cystine desulfurase system, consisting of the NADH-dependent cystine reductase (374) and the Plp-dependent cysteine desulfhydrase (75), is extraordinarily depressed or completely absent because of a deficiency in the two coenzymes (306, 487). Cysteine desulfhydrase is one of the enzymes most sensitive to B₆ deficiency (497). The activity of 5-HTP decarboxylase was also reduced in rat liver tumors. It decreased as soon as 2 weeks after feeding of the

cancerogenic compound 3,4-dimethylamino azobenzene. The normal liver tissue on the borders of the tumors had a normal enzymatic activity (266).

Just as the decrease in the B₆ content of tumors is not specific, but rather includes other vitamins, particularly those of the B group (for review see 187), the activity of various enzymes whose coenzyme is not Plp, *e.g.*, monoamine oxidase, is decreased in tumor tissue (266). The cause of tumor growth inhibition in B₆ deficiency, however, is not, as one might assume, due to an inhibition of transaminases. Against such a relationship would speak, for instance, the fact that cortisone causes marked growth inhibition of Walker carcinosarcoma and of other tumors in rats, but at the same time activates the glutamate-pyruvate transaminase to 14 times normal values (437). On the other hand, DOCA promotes tumor growth and inhibits transaminases (437).

The inhibition of tumor growth in B₆ deficiency is not necessarily caused by a decrease in the activity of vitamin B₆-dependent enzymes. It can also result from a deficit in growth factors for whose formation B₆ is necessary. Thus, the liver content of pantothenic acid, nicotinamide, and riboflavin (495), as well as NAD (546) and coenzyme A (547), falls in B₆ deficiency. This can be partially explained by an inhibition of synthesis. Indeed, coenzyme A synthesis in liver homogenates from B₆-deficient animals is inhibited and can be restored by the addition of Plp (561). Also, NAD synthesis is impaired since the injection of nicotinamide or nicotinic acid leads to significantly lower increases in NAD values in B₆-deficient animals than in normal animals (546).

Likewise the synthesis of purines, pyrimidines, flavines, and pterines depends on Plp-catalyzed reactions. The carbon and nitrogen atoms of these compounds are chiefly supplied by aspartic acid and glycine (for review see Braunstein, 75). Finally, it may be mentioned that the intestinal absorption of vitamin B₁₂ is diminished in B₆ deficiency so that serum and liver concentrations of B₁₂ are lowered (234, 416).

IX. VITAMIN B₆ AND HEMATOPOIESIS

Anemia can be a result of B₆ deficiency (117, 156, 334, 347, 401, 490, 552) in animals. In B₆-deficient rats blood regeneration following hemorrhage is retarded (464). Snyderman *et al.* (479) reported two children with micro- and hydrocephaly, in whom convulsions and anemia appeared during a B₆-deficient dietary regime.

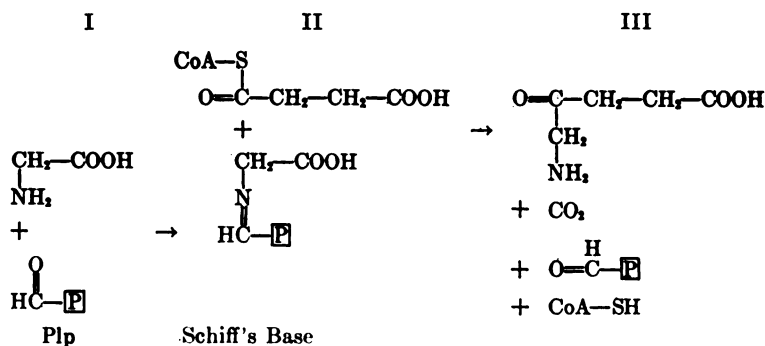
Genuine B₆-deficiency anemia in man was first observed by Harris *et al.* in 1956 (198). This rare clinical syndrome—only 17 cases have been described (122, 168)—agrees hematologically with the experimentally induced anemia in B₆-deficient animals: a microcytic, hypochromic anemia with an accompanying, strongly elevated serum iron level. The hematological picture in the human cases is similar to that of sidero-achrestic anemia (209), a disease of the erythropoietic system, sometimes hereditary, but usually acquired, appearing in older people, associated with a defect in the incorporation of iron into the hemoglobin molecule. The B₆ anemia is characterized by the appearance of sideroblasts and erythroblasts in bone marrow, high serum iron level, and liver hemosiderosis.

Many cases respond specifically to pyridoxine. Fe, Co, vitamin B₁₂, folic acid,

liver extract, ACTH, and corticoids, as well as androgens, are usually ineffective (122, 168, 170, 282). Blood transfusions have only a short-lasting therapeutic effect. Pyridoxine sometimes causes a drastic reticulocytosis as well as a progressive increase in the erythrocyte count and hemoglobin content. At the same time there is a disappearance of pathological cell forms and a fall in the serum iron concentration (170, 282). Pyridoxine alone is not always sufficient for a complete restitution of the hematological disturbance. Liver extracts, yeast extracts, and testosterone can act as adjuvants (122, 168, 231). There seems to be a familial frequency in this B₆-dependent anemia which occurs chiefly in mature individuals (65a). A relationship to Blackfan-Diamond syndrome has been suggested (9, 168).

Recently Kohn *et al.* (282) observed a fulminating anemia after INH treatment superimposed upon an already present mild hypochromic anemia. The serum transaminase activity, taken as a criterion of B₆ deficiency, was reduced to zero. Withdrawal of INH and the institution of B₆ therapy led to a reticulocytosis, increase in hemoglobin, a fall in the serum iron to subnormal values and an increase in the serum transaminase titers. The frequency of the occurrence of anemia during INH treatment deviates extraordinarily between 4 and 34% (354).

Essentially, a defect in heme formation is responsible for the appearance of the hypochromic anemia seen in B₆ deficiency, since B₆ is necessary for the synthesis of protoporphyrins on the one hand and probably for the incorporation of iron on the other. Plp is necessary as a coenzyme for δ -aminolevulinic acid synthetase (see section I F), which condenses glycine (I) and succinate (succinyl-Co A, II) to form δ -aminolevulinic acid (ALA, III). Two molecules of ALA further react by way of porphobilinogen and coproporphyrin to form the iron-free protoporphyrin.



Cartwright and Wintrobe (90), and numerous later authors (175, 176, 261, 373) assumed that the cause of B₆-deficiency anemia in pigs was the result of a disturbed protoporphyrin synthesis because they found that the protoporphyrin content of the erythrocytes was lowered. One can indeed show that heme formation from glycine and succinate in erythrocytes from B₆-deficient animals is inhibited and can be reactivated *in vitro* by the addition of Plp (457). Carbonyl reagents such as D-penicillamine, cysteine, and cyanide also inhibit the ALA-synthesis from glycine and succinate (175). However, a reduced protoporphyrin

content in the erythrocytes of the human anemia is not regularly found (90); the concentration of protoporphyrin was usually normal or even raised (168, 282). The same was true for the excretion of ALA in the urine: it could be normal (168) or even elevated (507).

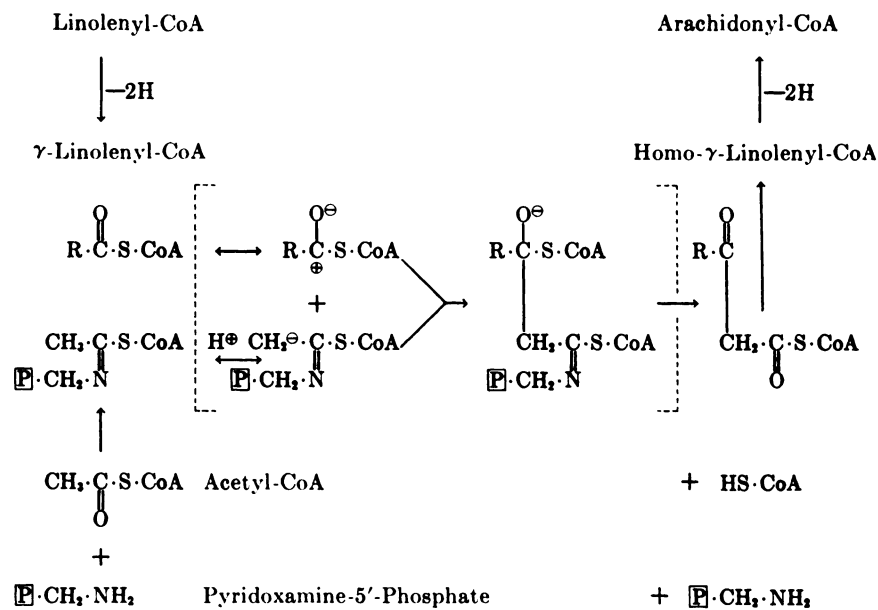
There are other reasons why it is unlikely that the only cause of human B₆-deficiency anemia is a disturbance of protoporphyrin synthesis. Heilmeyer and Clotten (208) could not restore the reduced heme synthesis in erythrocytes by the addition of ALA. In ducks the anemia in B₆ deficiency was not prevented by the injection of ALA (457). The explanation may be that B₆, in addition to its coenzyme function in ALA synthesis, is necessary for the incorporation of iron into the protoporphyrin molecule. Gardner and Nathan (168) found that in spite of normal blood concentrations of B₁₂ and folic acid, there was a strong inhibition of Fe⁵⁹ incorporation into the heme of erythrocytes in a patient with a pyridoxine-sensitive anemia. Yamada and Ogawa (560) have reported decreased iron incorporation in bone marrow cultures from B₆-deficient animals. This can also be produced by cycloserine, deoxypyridoxine, or INH and is improved by Plp.

Vitamin B₆, which is necessary for growth in animals and microorganisms, appears to be of importance also for the new formation of blood cells. This is indicated by its therapeutic action in toxic leucopenia (for review see 11, 363, 439).

X. VITAMIN B₆ AND METABOLISM OF FATTY ACIDS

In 1936, Birch (62) and György (193) reported that lipids with a high content of polyunsaturated fatty acids—designated as “vitamin F” at that time—were able to reduce acrodynia in vitamin B₆-deficient rats. Later in 1942, Quackenbush *et al.* (412) obtained similar results: linoleic acid and vitamin B₆ had an additive effect on acrodynia in rats. Further light on the mechanism of this action is shed by the finding of Witten and Holman (553) that vitamin B₆ is essential for the conversion of linoleate into the more unsaturated tetraenoic arachidonate. This agrees with the fact that in B₆-deficient rats the serum content of tetraenoic acids is decreased (184), and that supplementation of the diet with peanut oil in B₆-deficient chickens leads to a much more pronounced increase in linoleate than in arachidonate, as compared with control animals (115). The importance of vitamin B₆ for the utilization of linoleate is further documented by the finding that in B₆ deficiency, with or without supplementation with linoleic acid, the content of linoleate particularly in the phospholipid fraction of the liver is increased while concomitantly arachidonate is decreased (12, 450, 491, 548).

The mechanism of the action of vitamin B₆ might possibly be explained by *in vitro* experiments of Wakil (520) using liver mitochondria; pyridoxal- and pyridoxamine-phosphate appeared to be cofactors in the elongation of the fatty acid chain by incorporation of acetyl CoA. Pyridoxamine phosphate has been suggested to form a Schiff's base with acetyl CoA, which may activate the methyl group for the incorporation into the fatty acid molecule as shown in the following scheme adapted from Wakil (520).



It must be emphasized, however, that there are distinct differences in the histological manifestation of changes in the skin due to essential fatty acid deficiency and B₆ deficiency. The edema associated with essential fatty acid deficiency ("Burr-Burr syndrome") is predominantly localized in the ventral trunk skin, but edema caused by B₆ deficiency is confined to the dorsal skin of the hind paws. The tissue content of vitamin B₆ is normal in fatty acid deficiency (for review see Aaes-Jørgensen, 1).

The dermatitis seborrheica sicca often seen in B₆-deficient human beings responded favorably to oral or local therapy with vitamin B₆ as well as to high doses of linoleic acid (510), but the concomitant glossitis and neuritis did not respond to linoleic acid.

The weight loss in B₆ deficiency is essentially due to the mobilization of fat depots (48). In spite of a dietary supplementation with corn oil, B₆-deficient animals were unable to deposit fat, although they could still utilize carbohydrates (in contrast to protein) for the synthesis of fat (48, 127, 464).

The relationship between B₆ deficiency and atherosclerosis is still obscure. Rinehart and Greenberg (426) found in B₆-deficient monkeys vast atherosclerotic lesions in the abdominal aorta, and the iliac, femoral, renal, and coronary arteries, manifested by swelling of the intima with deposition of mucoids, and in more advanced stages by fibrous plaques without deposition of lipids in the media. While these findings were confirmed by Mushett and Emerson (367, 368) in monkeys and dogs, Olson (384) observed in B₆-deficient rats hypertension but no atherosclerotic changes of the vessels. Several authors reported hypercholesterolemia in rats, chickens, and monkeys, if B₆ deficiency was accompanied by atheromatous changes and deposition of cholesterol in the aorta (115, 184, 186). Since, according to Shah *et al.* (462), cholesterol synthesis is not

accelerated in B₆ deficiency, hypercholesterolemia rather seems to be due to an increased mobilization of cholesterol in the liver (184).

The relationship between human atherosclerosis and B₆ deficiency is controversial (for review see 439, 528). A therapeutic effect of vitamin B₆ appears doubtful (439), and only occasionally can an increased urinary excretion of xanthurenic acid be observed in atherosclerotic patients (528), whereas according to Boxer *et al.* (70), the Plp content of the leukocytes is sometimes significantly decreased.

XI. RADIOPROTECTIVE ACTION OF VITAMIN B₆

In 1948, Goldfeder *et al.* (180) reported that pretreatment with pyridoxine prolonged the survival time of mice exposed to 350 r radiation, from 14 days to 100 days. Bridges and Koch and others (77, 280, 297) could not confirm these results with pyridoxine, but found that Plp, injected 15 minutes before the radiation, did protect. On the other hand, radiation damage is not more severe in the B₆-deficient rat, and the B₆ content of the liver is not reduced by radiation (339). Also, radiation sensitivity of mice is not increased by pretreatment with INH or deoxypyridoxine; it is increased by penicillamine (280, 296). In the monkey, radiation caused only a brief increase in the B₆ excretion (383). In the rat after a 525 r radiation dose, there was a strong inhibition of the absorption of pyridoxine from the intestine, which became maximal after 3 days and disappeared after 7 to 17 days (128).

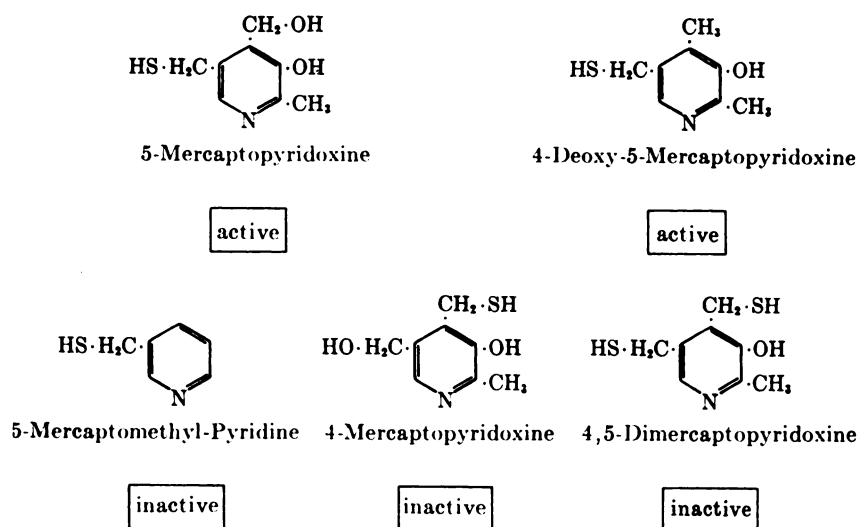
The increased excretion of xanthurenic acid in the rat and mouse could also be indicative of a radiation-induced B₆ deficiency (202, 301, 349). Following radiation, the xanthurenic acid excretion is first diminished for a short time and then increases sharply during the following 7 to 10 days, while the taurine excretion is first increased and later falls to subnormal values. At the same time the content of cysteine sulfinic acid in the spleen is increased, probably due to an inhibition of the B₆-dependent cysteine sulfinic acid decarboxylase (4, 301, 351). Other intermediary products of tryptophan metabolism, such as kynurenine, kynurenic acid, and anthranilic acid, are also excreted in increased amounts. The cause of this is probably a radiation-induced activation of hepatic tryptophan pyrrolase which results in overloading of the capacity of the B₆-dependent kynureninase by the increased amounts of metabolites (496). *In vitro*, the different forms of vitamin B₆ are rapidly destroyed by irradiation (330, 370).

The fact that serotonin and to a lesser extent 5-HTP as well as ATP exert a radioprotective action also could speak for a relationship between B₆ enzymes and radiation damage (297, 298, 300). The relatively weak action of 5-HTP is potentiated to approximately that of serotonin when it is injected along with Plp and ATP 15 minutes before irradiation. It appears that ionizing radiation inhibits decarboxylation of 5-HTP, possibly because of an ATP deficiency (297, 298). Reserpine also has a protective effect when administered 12 hours before irradiation (299, 350).

Vitamin B₆ not only hinders the lethal action of high doses of radiation in animals, but also diminishes the toxic action of sublethal doses in man. Most

radiologists believe that the clinical symptoms of radiation sickness, *e.g.*, nausea and vomiting, can be favorably influenced by B₆. Maxfield *et al.* (331) could bring about the disappearance of the radiation-induced vomiting and nausea within 30 minutes by the administration of 25 mg of pyridoxine hydrochloride. Similar results have been reported by other authors (173, 196, 385). The usual appearance of nausea during 600 r radiation therapy 5 times a week could be suppressed by the intravenous injection of 25 mg of pyridoxine. Withdrawal of pyridoxine immediately led to the recurrence of nausea and vomiting. The accompanying diarrhea was not influenced by B₆ (459). However, Silverman *et al.* (467) denied any curative effect against deep radiation in a comparison of pyridoxine, Dramamine (dimenhydrinate), and placebo.

The radioprotective action of sulfur-containing B₆ analogues (without anti-metabolic effects) reported in the last few years by Langendorff *et al.* (278, 281, 296) must be placed beside the well-known protection exerted by sulfur-containing compounds of the cysteine-cysteinamine group. 2-Methyl-3-hydroxy-4-hydroxymethyl-5-mercaptomethyl-pyridine was active, even when administered as late as 2 hours after irradiation. The protective action is probably less the result of the B₆ skeleton, than of the SH-group found in position 5, as shown by the following formulae:



Although nothing is known about the mechanism of the radioprotective effect of vitamin B₆, particularly of its antiemetic action, a comparison with its therapeutic effect in hyperemesis gravidarum appears feasible (for review see 11, 439). In both cases there is an increase in protein breakdown and transformation with an augmented influx of protein degradation products, such as peptides and amino acids, into the circulation, and on the other hand a relative B₆ deficiency. This may explain why vitamin B₆ has an antiemetic action only in radiation sickness and in pregnancy. Although its use is often recommended in cases of motion sickness, Chinn *et al.* (97) were unable to demonstrate any beneficial effect

of vitamin B₆ in extensive investigations on a great number of persons. Pyridoxine proved to be ineffective also in experimental vomiting provoked in dogs by application of apomorphine (453).

XII. PHARMACOLOGY AND TOXICOLOGY OF VITAMIN B₆ AND ITS DERIVATIVES

As a vitamin, B₆ is active even in small doses in B₆ deficiency. In B₆-saturated organisms, however, effects can be demonstrated only with high doses. It seems doubtful whether these effects are actually due to its vitamin character, *i.e.*, to an alteration of B₆-dependent biochemical reactions (285).

The toxicity (1/LD₅₀) of the three naturally occurring forms of vitamin B₆ (pyridoxine, pyridoxamine, pyridoxal), as well as of pyridoxic acid, is much higher after intravenous than after subcutaneous or oral application. Pyridoxal is 2 to 5 times more toxic than pyridoxine or pyridoxamine (279, 285, 505). Replacement of the hydroxyl groups by sulfhydryl groups increases the toxicity (279, 285), while a still more pronounced increase in toxicity can be achieved by methylating the hydroxymethyl group in position 4 (R—CH₂—OH → R—CH₂—OCH₃): the product, 4'-methyl-pyridoxine, is an antimetabolite to vitamin B₆ (505).

The toxic effects of high doses of vitamin B₆ are mainly of central origin: impaired postural reflexes, tachypnea, convulsions, and paralyses; in addition, salivation and vomiting occur in cats and pigeons (279, 285, 505). Degeneration of the dorsal columns and the dorsal roots of the spinal cord, of ganglia and sensory nerves were observed postmortem in dogs and cats which had received 2 to 6 g of pyridoxine per day for a long period of time (14). Following the administration of 200 mg of pyridoxine to cats for 20 days Astruc (22) found changes of the adrenals resembling those after nonspecific stress.

It is of interest to note that a number of signs of B₆ hypervitaminosis are similar to those found in B₆ deficiency. This might possibly be explained by the observation that following the administration of large amounts of pyridoxine, pyridoxamine, or pyridoxal in mice, the pyridoxamine- and pyridoxal-phosphate content in the central nervous system was significantly lowered (28).

In cats, 10 mg of pyridoxine had a slight sedative effect if given intracisterially, while 20 mg caused excitation after a transient period of sedation; 3 mg of pyridoxal or 0.3 mg of pyridoxal-phosphate were followed by marked excitation, salivation and short-lasting convulsive seizures. Pyridoxamine caused sedation, whereas pyridoxamine-phosphate had no effect (32, 285). In human beings pyridoxine is said to exhibit sedative actions (484).

In contrast to the centrally excitatory 5-sulfhydryl derivatives of pyridoxine, 5-chloro-pyridoxine causes sleep-like sedation (279). Dipyridoxolyl-5'-disulfide and dipyridoxolyl-4'-disulfide, in contrast to pyridoxine, produced sedation and a prolongation of the spontaneous sleeping time, and caused blepharospasm and miosis (285, 364). Long-term administration of the disulfides and of pyridoxine to rhesus monkeys resulted in a reduction of the spontaneous motor activity and aggressiveness (364).

In Regnier's test with pyridoxine and several of its derivatives (*e.g.*, 4-pyri-

doxic acid, 4-deoxypyridoxine-5'-disulfide, and pyridoxamine-5'-disulfide) a centrally induced analgesia is demonstrable. This seems to be a direct action not related to any coenzyme function, since these compounds, with the exception of pyridoxine itself, have little or no anticonvulsive properties against seizures induced by toxopyrimidine (285).

In isolated organs (ileum, uterus), pyridoxine and its derivatives had no characteristic effects (for review see Unna, 505). Since the chronotropic and inotropic effects on isolated rabbit atria as described by Levine *et al.* (307) are associated with an increased uptake of sodium and a decreased release of potassium, they could perhaps be explained by an alteration of the active transport of ions (see section VII).

Pyridoxine itself has no effect on the blood pressure if given intravenously. However, it potentiated the blood-pressure-raising effect of L-dopa, as shown in cats (31) and in man (124). This effect is due to its function as a coenzyme for dopa decarboxylase.

CONCLUDING REMARKS

In the present contribution we have attempted to demonstrate that vitamin B₆, particularly its catalytically most important form pyridoxal-5'-phosphate (Plp), is a many-sided biocatalyst by virtue of its high chemical reactivity. For the same reason it is particularly susceptible to chemical inactivation. As with other vitamins, a *chronically* insufficient intake of vitamin B₆ results in typical signs of an avitaminosis in man and animals. Moreover, many drugs exert acute effects by inactivation of Plp, thus interfering with B₆-catalyzed biochemical reactions. Therefore, vitamin B₆ has more "pharmacological aspects" than other vitamins.

1. Even a short-lasting functional inactivation of Plp by carbonyl reagents, *e.g.*, semicarbazide, thiocarbohydrazide, and others, leading to the formation of hydrazones by reacting with the aldehyde group of Plp, promptly induces convulsions within minutes after their application, this result indicating a functional avitaminosis of the brain. Convulsive carbonyl reagents, such as hydrazine and hydroxylamine, are widely used industrially, and they may also be found among drugs used in man. Thus, the tuberculostatic agent isonicotinylhydrazide (INH) directly reacts with Plp, whereas cycloserine can do so only after biotransformation to a hydroxylamine derivative. While single application of a high dose of hydrazides provokes acute cerebral convulsions, chronic administration, even in "therapeutic" doses, can lead to a widespread neuritis in the peripheral nervous system. It must be admitted, however, that the intimate mechanism of these actions has not been established clearly despite many efforts. The reasons may be found in the extraordinarily large number of reactions in which Plp serves as coenzyme. In addition, all these reactions are susceptible to inhibition to a highly varying degree, depending upon the differing affinities of the coenzyme to the apoenzymes. Thus, convulsive doses of hydrazides inhibit the brain-specific, Plp-dependent glutamic acid-decarboxylase, thereby leading to a decrease of the γ -aminobutyric acid (GABA) in the brain; at the

same time, however, the activity of the equally Plp-dependent aromatic amino acid decarboxylase is unchanged and the amine content of brain remains normal. On the other hand, there are carbonyl reagents, *e.g.*, hydroxylamine and hydrazine, which induce convulsions, although they cause an elevation of brain GABA. These compounds apparently inhibit the GABA- α -ketoglutaric acid transaminase to a higher degree than the glutamic acid decarboxylase. From this it follows that there is no causal relationship between the GABA content of the brain and the susceptibility of cerebral motor centers to convulsive stimuli; more likely the lowered turnover in the glutamic acid-GABA pathway is the decisive factor because of its importance for the oxidative metabolism of the brain. However, it still remains to be established whether an impairment of the GABA pathway is of importance at all for the induction of convulsions, or whether the detectable biochemical alterations are secondary to an unknown process. In this connection it must be emphasized that vitamin B₆ has a protective action against convulsions due to B₆ deficiency only, but is devoid of any therapeutic effect in the treatment of human epilepsy.

As to the neuritic symptoms caused by long-term administration even of therapeutic doses of hydrazides, *e.g.*, INH, it should be realized that Plp is the coenzyme of the biosynthesis of sphingosine and hence of sphingomyelin, an essential constituent of nervous tissue.

2. Some of the most important B₆-catalyzed reactions are involved in the biosynthesis of amino acids from nitrogen-free precursors, and therefore in the synthesis of protein. B₆ is also involved in the intestinal absorption and in the active transport of amino acids. From this results its importance for growth and cell replacement. Tumor growth can be inhibited by dietary as well as by chemically induced B₆ deficiency. The function of vitamin B₆ in amino acid and protein metabolism might also be the base for its antiemetic action in cases of hyperemesis gravidarum and radiation sickness, since in both instances we are dealing with alterations in the protein metabolism with an abnormally high B₆ requirement.

3. It is of interest that impairment of vitamin B₆-catalyzed metabolic reactions can be the mechanism underlying the toxic action of drugs, *e.g.*, the convulsive hydrazides. On the other hand, therapeutic drug effects, *e.g.*, the antihypertensive action of α -methyldopa, may be mediated by vitamin B₆ through its coenzyme function.

4. B₆ is also necessary for the synthesis of other biocatalysts in the organism. We have pointed out that the synthesis of heme is specifically B₆-dependent, since Plp is necessary for the formation of δ -aminolevulinic acid, the precursor of pyrroles and porphyrins. Therefore, anemia is a specific symptom of B₆ deficiency, while rather nonspecific signs can arise from the fact that B₆ is also essential for the synthesis of numerous enzymic cellular heme derivatives. Other rather nonspecific signs of B₆ deficiency could be the result of a decreased synthesis of NAD and coenzyme A, or of an impaired intestinal absorption of vitamin B₁₂. For these reasons it is difficult or impossible satisfactorily to interpret manifestations resulting from a B₆ deficiency in biochemical terms.

5. Since vitamin B₆, like other vitamins, is not a pharmacon but a physiological biocatalyst, it follows that specific therapeutic effects from its administration can be expected only in cases of a lack of this compound, as for instance in babies fed with B₆-poor milk preparations or in cases of vitamin B₆-dependence, where B₆ is the specific anticonvulsant. Also the administration of vitamin B₆ in arteriosclerosis could be to a certain degree a treatment directed against the etiology of the disease since this vitamin is involved in the enzymatic biosynthesis of unsaturated fatty acids. The same could be true for the treatment of Parkinsonians with vitamin B₆ because of its coenzyme function in the synthesis of dopamine. The clinical aspects of vitamin B₆ discussed in our contribution were confined to diseases in which causal relationships between pathogenesis and a disturbance of vitamin B₆-catalyzed reactions could be assumed.

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